

The distribution of Fe across the shelf of the Western Antarctic Peninsula at the start of the phytoplankton growing season

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ABSTRACT

The Western Antarctic Peninsula (WAP) is a rapidly changing region with receding sea-ice cover. This region generally has increased phytoplankton productivity on the continental slope with a decrease observed in off-shelf waters located further northward. This study aims to improve the understanding of the distribution of iron (Fe) in this climatically important oceanic region during early sea-ice retreat, as well as the impact of receding ice cover on Fe concentrations and the importance of Fe and its sources at the beginning of the phytoplankton growth season. Five ocean transects were sampled along the Palmer Long Term Ecological Research (Pal-LTER) grid with an additional oceanward extension to access off-shelf waters during the austral spring (October and November) of 2014. High Fe inputs into the entire water column from ice melt as well as from sediments increased dissolved Fe (DFe) and total dissolvable Fe (TDFe) concentrations across the shelf (off-shelf DFe: 0.38 ± 0.30 nmol/L and TDFe: 2.23 ± 2.95 nmol/L versus shelf DFe: 1.54 ± 1.38 nmol/L and TDFe: 19.47 ± 23.82 nmol/L). The combination of meteoric meltwater and shallow sedimentary sources is strongest over the shelf and increased landward towards the WAP. Additionally, a winter sea-ice formation signature was detected in inner shelf waters that appeared to contribute to DFe concentrations. Relatively warm Circumpolar Deep Water (CDW) that flows onto the shelf through troughs is likely modified by non-reductive sedimentary input of Fe into the water column. The increase in Fe concentrations in WAP waters in early spring could trigger enhanced phytoplankton productivity across the shelf, although the highest productivity levels were observed in off-shelf waters, likely related to improved light conditions following receding sea-ice cover. The relatively high productivity levels in WAP off-shelf waters are presumably caused by two factors: (1) supply from below, and (2) the transport of Fe from the shelf to the off-shelf region. However, 80–90% of Fe introduced into shelf waters was removed from the upper water column prior to reaching off-shelf waters, reducing the influence of the coastal Fe source. Low concentrations of DFe (0.24 ± 0.26 nmol/L) in off-shelf waters (upper 100 m) of the WAP coincide with relatively elevated chlorophyll *a* concentrations (0.66 ± 0.56 µg/L), implying that uptake of available DFe had already occurred prior to sampling. Our results imply that the horizontal supply of DFe from the shelf as well as total DFe supply into the surface mixed layer might not be sufficiently high to support productivity in the off-shelf waters of the WAP, underlining the development of an Fe deficit early in the growth season.

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1. Introduction

The Western Antarctic Peninsula (WAP) is one of the most intensively studied regions of Antarctica due to the Palmer Long Term Ecological Research (Pal-LTER) program that began in 1990 (Ducklow et al., 2013; Steinberg et al., 2012). This region is known to have high primary productivity (Vernet et al., 2008) in some coastal regions, known as biological ‘hotspots’ (Sherrell et al., 2018), featuring abundant organisms at all trophic levels (Kavanaugh et al., 2015; Schofield et al., 2013). The regional productivity is influenced by the extent, duration, and seasonality of sea-ice and glacial discharge (Ducklow et al., 2013). Marine primary productivity results in carbon drawdown from the atmosphere with subsequent transfer of organic matter to the deep ocean via the biological pump, establishing the Southern Ocean as a potential sequestration region for carbon dioxide (CO₂) (Brown et al., 2019; Henley et al., 2020; Martin, 1990; Yang et al., 2021). The continental shelf of Antarctica accounts for ~10% of Southern Ocean productivity with the remaining ~90% occurring in off-shelf and open ocean waters (Arrigo et al., 2008), although the productivity per unit of surface area is much larger in shelf regions. The Antarctic Peninsula experienced rapid warming in the second half of the 20th century (Marshall et al., 2006), resulting in surface melt of glaciers (Barrand et al., 2013; Holland et al., 2015; Kunz et al., 2012; Pritchard and Vaughan, 2007). More importantly, intruding warm deep waters along the WAP promote basal melt that further decreases ice sheet cover (Cook et al., 2016).

Phytoplankton require major nutrients and micronutrients, including trace metals, to grow. A critically important micronutrient is iron (Fe), which acts as a primary electron contributor and acceptor during photosynthesis and as a cofactor in enzymes in nitrate assimilation (Raven, 1988). Additionally, due to the different requirements of different species, dissolved iron (DFe) concentrations in seawater can control the primary production and the species composition of the phytoplankton community (Coale et al., 1996). This is especially the case in ‘high-nutrient, low-chlorophyll (HNLC)’ waters, such as those of the Southern Ocean south of the Polar Front (de Baar, 2005; Henley et al., 2020; Tagliabue et al., 2017), which have elevated surface water macronutrient concentrations but relatively low levels of primary production due to the combination of low levels of light and biologically important trace metals, notably Fe (Moore, 2013). It has been suggested previously that Fe played a crucial role in maintaining low levels of atmospheric CO₂ during the Last Glacial Maximum by stimulating high marine productivity (Martin, 1990), making Fe a crucial player in global climate.

Dissolved Fe is known to have a hybrid type distribution in the oceans that is dominated by microbial uptake, recycling and scavenging processes, leading to generally low surface concentrations (Bruland et al., 2014). Input processes of dissolved and particulate Fe in the Southern Ocean include a combination of: (1) atmospheric deposition (Gao et al., 2020; Gao et al., 2013; Tagliabue et al., 2009; Winton et al., 2015), (2) hydrothermal input (Tagliabue et al., 2010), (3) meltwater from sea-ice (Lannuzel et al., 2011; Lannuzel et al., 2007, 2010), glaciers and icebergs (Annett et al., 2015; Bown et al., 2017; Lin et al., 2011; Raiswell et al., 2016; Raiswell et al., 2008), (4) sea-ice formation (Schallenberg et al., 2016), (5) sedimentary input from the resuspension of sedimentary particles (de Jong, 2012), (6) dissolution from sediments and/or pore waters (de Jong, 2012; Hatta et al., 2013; Measures et al., 2013), (7) mixing processes such as advection and diffusion from distant and/or nearby sources (De Jong et al., 2015; Tagliabue et al., 2014) and (8) buoyancy-driven circulation due to melting ice via ‘the meltwater pump’ that results in the horizontal and vertical supply of Fe derived from both the sediments as well as glacial melt (St-Laurent et al., 2017; St-Laurent et al., 2019; Twelves et al., 2021). The latter process, where glacial melt induced by warm intruding water leads to upwelling, has been shown to be especially important for delivering Fe to surface waters in some Antarctic shelf seas (e.g. St-Laurent et al., 2017). Removal processes include (1) oxidative precipitation (Millero et al., 1987), (2)

adsorptive scavenging (Measures et al., 2013), and (3) phytoplankton uptake (Bown et al., 2017).

Previous studies have based their findings of the processes controlling the distribution of Fe in WAP waters on surface Fe concentrations (~2 m depth) across the shelf (Annett et al., 2017) and a limited area in the inner-shelf (Sherrell et al., 2018) during the austral summer. These investigations provided important initial constraints on some of the major processes influencing Fe distributions in WAP waters, as well as in other shelf regions of Antarctica. The Fe concentrations in WAP waters have been suggested to be controlled by inputs from meltwater, (shallow) sedimentary sources across the shelf, and horizontal transport of CDW waters across the shelf break and advection of coastal waters from the inner shelf regions (Annett et al., 2017; Sherrell et al., 2018). However, summer Fe concentrations are very much depleted, making it harder to assess the impact of other essential processes on Fe distributions in the surface layer. Our study builds upon these earlier findings by reporting Fe distributions and concentrations in waters adjacent to the WAP based on depth profiles across the WAP shelf, from near the coast to the off-shelf region. Moreover, this study reports Fe distributions for the beginning of the phytoplankton growth season (early austral spring), allowing the Fe stock at the beginning of the growth season, when sea-ice is retreating, to be determined. In addition, Mn values are reported and used here to help identify sedimentary sources of Fe, while the processes affecting Mn distributions and those of other essential trace metals will be discussed elsewhere (in preparation). Our study aims to differentiate and characterize the sources of Fe in the shelf region and to assess the influence of shelf waters as a potential Fe source to the off-shelf region.

2. Methods

2.1. Study region and hydrography

The study region comprises the Pal-LTER grid with an ocean-ward extension (Fig. 1A). This grid is located on the western side of the Antarctic Peninsula. The Pal-LTER sampling transects are arranged perpendicular to the peninsula at 100 km intervals along lines 200, 300, 400, 600 and 700, extending from the coastline to 200 km offshore (Fig. 1A). In this study, the lines were extended by an additional 200 km offshore from the shelf break to include off-shelf samples as previously described (Arrigo et al., 2017).

The offshore region of the study area is characterized by cold, fresh Antarctic Surface Water (AASW) underlain by warmer Upper Circumpolar Deep Water (UCDW) that is characterized by an oxygen minimum, and Lower Circumpolar Deep Water (LCDW) that is slightly colder but more saline than UCDW (Fig. 1 B–C, S1). The AASW is defined as water with a potential temperature < -1.4 °C and salinity <34, while UCDW consists of water with a potential temperature > 1.5 °C and salinity between 34.4 and 34.7, and LCDW is defined as water with a potential temperature between 1.0 and 1.5 °C and salinity >34.7 following earlier definitions (Arrigo et al., 2017).

The waters adjacent to the WAP contain the Polar Front (PF), southern ACC Front (SACCF) and the southern boundary of the ACC (SBACC). The Polar Front is located north of the study area, whereas the SACCF and SBACC are located within the study region (Fig. 1A). Stations between these fronts have similar upper water column characteristics, whereas stations separated by the fronts have different characteristics, becoming warmer, fresher and less dense towards the off-shelf. The CDW water from the ACC intrudes across the shelf through troughs, with eddy events, flooding over the shelf (Moffat and Meredith, 2018).

The study region also contains two clockwise gyres (Hofmann et al., 1996), one near Anvers and Brabant Islands and the other near Adelaide Island. These two gyres are the result of a northeast flowing current along the shelf edge following the ACC route and the southwest flowing Antarctic Peninsula Coastal Current (APCC) (Moffat et al., 2008). The APCC is defined as a strong, narrow current near the coast formed by

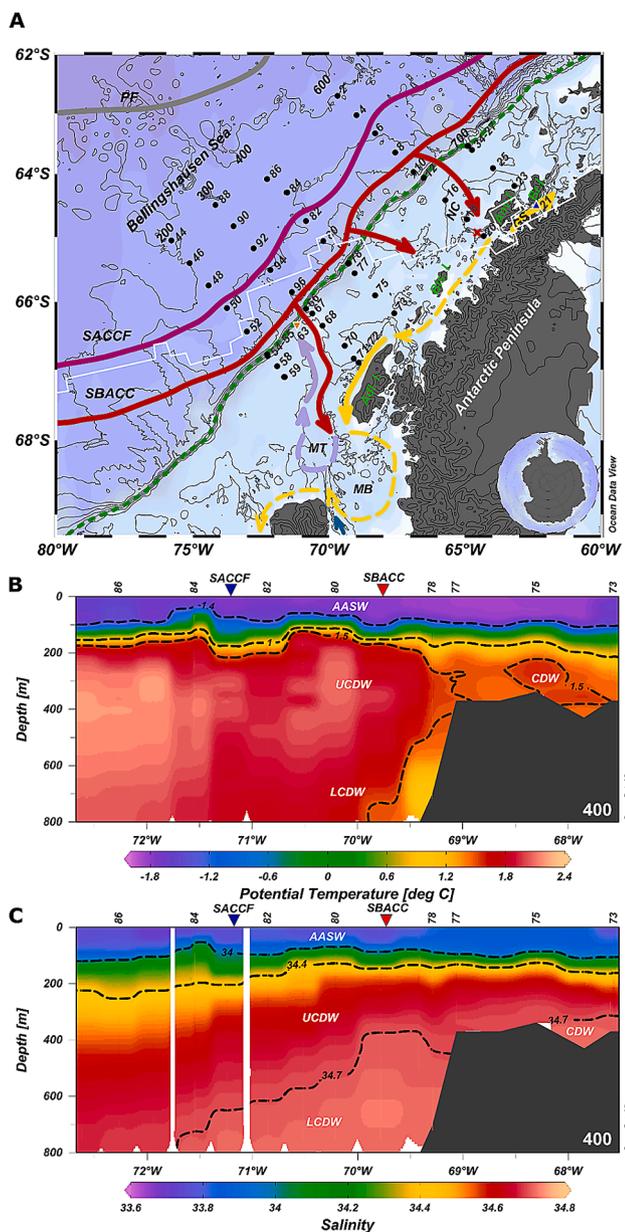


Fig. 1. A. Map of the Western Antarctic Peninsula (WAP) study region with sampling lines 200 to 700 and station numbers indicated next to the station symbols. **B.** Distribution of potential temperature in color scale along sampling line 400 with water masses indicated with black dashed lines and the location of fronts indicated by a blue triangle for SACCF and by a red triangle for SBACC. **C.** Distribution of salinity in color scale along sampling line 400 with AASW, UCDW, LCDW indicated with black dashed lines and the location of fronts indicated by a blue triangle for SACCF and by a red triangle for SBACC. Abbreviations: Adl. I. – Adelaide island; Anv. I. – Anvers Island; AASW – Antarctic Surface Water, UCDW – Upper Circumpolar Deep Water, LCDW – Lower Circumpolar Deep Water, Bis. I. – Biscoe Islands; Br. I. – Brabant Island; GS – Gerlache Strait; MB – Marguerite Bay; MT – Marguerite Trough; NC – Northern Canyon; PF – Polar Front; SACCF – Southern Antarctic Circumpolar Current Front; SBACC – Southern Boundary of Antarctic Circumpolar Current. Arrows: Yellow arrow – Antarctic Polar Coastal Current (APCC: solid – confirmed route, and dashed – suspected route); Red arrows – intrusion of CDW; Purple arrow – outflow current (solid – confirmed current, and dashed – suspected current); Cyan arrow – suspected George VI Ice Shelf input; Red cross symbol – Palmer Deep Trough; White-blue triangle symbol – station 21; White-orange triangle symbol – station 63; Dashed green line – 500 m isobaths shelf break; White solid line – sea-ice extent during the sampling period. Adapted from Moffat and Meredith, 2018. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

freshwater discharge and downwelling-favorable winds (Moffat et al., 2008). This current flows southwest from Anvers Island, along Biscoe and Adelaide Islands into Marguerite Bay, south of Adelaide Island, and northeast between the Antarctic Peninsula and Anvers and Brabant Islands (Moffat et al., 2008; Moffat and Meredith, 2018; Savidge and Amft, 2009) (Fig. 1A). The APCC current can reach depths of 100–150 m and be up to 20 km wide (Moffat and Owens, 2008; Moffat and Meredith, 2018; Savidge and Amft, 2009).

Samples were obtained in early spring (October and November 2014) while melting of the seasonal sea-ice was ongoing. The mean open water area was ~170,000 km² in August/September and increased to ~332,000 km² in December/January (Arrigo et al., 2017). The peak in ice-free water was reached one month earlier (December) in off-shelf waters than in shelf waters (January) (Arrigo et al., 2017). Therefore, the peak in ice-free water occurred after the sampling period, but the extent of ice-free water was already larger in the off-shelf region and ice cover increased shoreward from the shelf break towards the continent (Arrigo et al., 2017). The southern zone of the sampling region (lines 200 and 300) had the highest sea-ice cover, approaching 100%, and in the northernmost part of the sampling region (line 700) cover was only 30% during the time of sampling (Arrigo et al., 2017).

2.2. Sample collection

Seawater samples were collected with a trace-metal clean CTD rosette equipped with GO-FLO bottles until the rosette was lost due to failure of the Kevlar cable after completing line 600 and half of line 700. Thereafter, seawater samples were collected with GO-FLO bottles deployed on the remaining Kevlar cable and triggered manually with messengers. Samples for trace metal concentration analysis, including both dissolved metals (DM) and total dissolvable metals (TDM), were collected into 60 mL LDPE (Nalgene) bottles that were rigorously acid-cleaned prior to the expedition following GEOTRACES protocols (Gerringa et al., 2020). Samples for DM analysis were filtered through a 0.2 µm filter (Sartobran 300, Sartorius) directly from the GO-FLO bottles under 0.5 bar nitrogen gas overpressure, while TDM samples were unfiltered. All samples were then acidified to 0.024 M HCl (pH ~ 1.8) with ultra-pure HCl (Optima grade Fisher Scientific). All samples were collected under clean conditions inside a trace metal clean van equipped with trace metal clean laminar flow benches. The labile particulate fraction of a metal (lpM) was calculated by subtracting DM from TDM concentrations.

Samples for δ¹⁸O analysis were collected into 25 mL glass vials and stored at room temperature. Samples for chlorophyll *a* analysis were collected as described previously (Arrigo et al., 2017). Samples for major nutrients (nitrate and phosphate) were analyzed simultaneously with a discrete autoanalyzer TRAACS 800 (Technicon) in the shore-based laboratory of the Royal Netherlands Institute for Sea Research (NIOZ), the Netherlands.

2.3. Sample analyses

2.3.1. Multi-element concentration analysis

All filtered samples and some unfiltered samples were analyzed at NIOZ, while the remaining unfiltered samples were analyzed at the University of Otago. Pre-concentration of seawater samples was done using a seaFAST pico (ESI, USA) system equipped with Nobias-chelate PA1 chelating resin (Hitachi High-technologies, Japan) after UV-oxidation as detailed by Gerringa et al. (2020). Samples were analyzed using a Thermo Element2 Sector-Field ICP-MS (SF ICP-MS) at NIOZ and a Nu Attom SF ICP-MS at the University of Otago. Using a microFAST system (ESI), 250 µL of sample was introduced into the ICP-MS via an ESI-PC3 Peltier cooled spray chamber (4 °C) with a microFAST PFA nebulizer at a constant rate of 35 µL min⁻¹. The preconcentration recoveries were verified in every analytical run by comparing the slope of the seawater calibration curve (multi-element standard added to

the seawater sample from the surface waters of South Pacific Ocean) and the eluent calibration curve (multi-element standard added directly to the elution acid) after Biller and Bruland (2012). Recoveries were > 98% for both Fe and Mn.

Blank contributions from sample handling, pre-concentration and analysis were determined by analyzing acidified high-purity water (~1.8 pH) (Milli-Q Element purification unit, Millipore Ltd., USA) as a sample. The determined blank values for both laboratories were comparable (Table 1).

The accuracy and precision of the measurements was determined by measuring the reference sample GEOTRACES South Pacific (GSP) as well as in-house reference seawater samples, notably a surface seawater sample (Munida 0) for low concentration samples, as well as North Atlantic Deep Water (NADW) and surface seawater spiked with known concentrations of trace metals (Munida +) for medium to high concentration samples (Table 1). Additionally, the NASS 7 certified reference material was analyzed and the concentrations of two stations (72 and 73) were determined in both laboratories (NIOZ and University of Otago) for intercomparison between the institutes, which showed good agreement within the magnitude of their analytical uncertainties (Table 1).

2.3.2. $\delta^{18}\text{O}$ analysis

Samples comprising 0.5 mL were analyzed at the Isotraces Laboratory, University of Otago by equilibrating with 12 mL of 0.3% CO_2 in helium on a Thermo (Bremen) GasBench preparation unit for 18 h at 25.0 ± 0.01 °C. Ten repeat injections of the equilibrated gas were measured with a Thermo Advantage isotope ratio mass spectrometer (IRMS) in continuous-flow mode. The 10 raw results were filtered by removal of values more than 1 standard deviation from the average. The filtered average was corrected to the international VSMOW-SLAP isotope scale using a three-point calibration provided by three laboratory standards analyzed before and after every batch of 84 samples. In addition, a control sample chosen to be similar to the samples being measured was analyzed at every 12th position to correct for instrumental drift. The instrumental drift correction, if applied, was calculated from a linear regression of the control sample results against time. Consensus values for the laboratory standards have been obtained from six-year internal laboratory calibration records against primary reference materials, VSMOW, GISP and SLAP, an external six member interlaboratory comparison exercise, and by back-calculation from the

~170 member IAEA interlaboratory comparison exercise, WICO2012. The laboratory standards and their consensus values are as follows: ICE ($\delta^{18}\text{O}_{\text{VSMOW}} = -32.097 \pm 0.075$ milliurey (mUr = 1‰)), TAP ($\delta^{18}\text{O}_{\text{VSMOW}} = -11.432 \pm 0.038$ mUr), and SEA ($\delta^{18}\text{O}_{\text{VSMOW}} = -0.029 \pm 0.036$ mUr).

The proportions of meteoric and seawater were calculated using the method of (Meredith et al., 2013). Three-component mass balance equations for meteoric water, ice melt and CDW were solved using the inverse matrix technique.

$$\begin{aligned} f_{\text{sim}} + f_{\text{met}} + f_{\text{CDW}} &= 1 \\ S_{\text{sim}} \times f_{\text{sim}} + S_{\text{met}} \times f_{\text{met}} + S_{\text{CDW}} \times f_{\text{CDW}} &= S \\ \delta_{\text{sim}} \times f_{\text{sim}} + \delta_{\text{met}} \times f_{\text{met}} + \delta_{\text{CDW}} \times f_{\text{CDW}} &= \delta \end{aligned} \quad (1)$$

where f_{sim} , f_{met} , and f_{CDW} are the fractions of sea ice melt, meteoric water, and CDW, respectively; S_{sim} , S_{met} , and S_{CDW} are the respective salinities of the pure components of sea ice melt, meteoric water, and CDW; and δ_{sim} , δ_{met} , and δ_{CDW} are their corresponding $\delta^{18}\text{O}$ values. The quantities S and δ are the measured values of salinity and $\delta^{18}\text{O}$ in each water sample for which the sea ice and meteoric freshwater contributions are being derived. Following the Meredith et al. (2013) study in the LTER grid region, the endmembers are assigned values of 34.73 and + 0.1 mUr for the salinity and $\delta^{18}\text{O}$ of the CDW endmember, respectively. Sea ice melt is assigned corresponding values of 7 and + 2.1 mUr, and meteoric water is assigned values of 0 and - 16 mUr.

2.4. Dissolved iron supply calculations

2.4.1. Horizontal diffusive supply of dissolved iron

The horizontal supply in the surface mixed layer (SML) and below the SML from the peninsula towards the off-shelf region was estimated for each sampling line using a one-dimensional advection/diffusion model simplified from de Jong, 2012 by only estimating the horizontal diffusive supply (J_{hs}).

$$J_{hs} = -K_h \times \left(\frac{\partial^2 [DFe]}{\partial x^2} \right) \quad (2)$$

Similar horizontal supplies via diffusion only were estimated in other parts of the Southern Ocean as well (Bucciarelli et al., 2001; Gerringa et al., 2015; Johnson et al., 1997; Planquette et al., 2007). The simplified one-dimensional advection/diffusion model equation uses only horizontal eddy diffusive flux divergence ($\mu\text{mol m}^{-3} \text{s}^{-1}$) in Eq. (2), which is

Table 1

Blanks, limits of detection and analytical performance of seawater trace metal consensus samples, reference materials and intercomparison stations.

| Elements | Blank | | | n | | | | |
|--------------------------------------|---------------|----|---------------|---------------|---------------|-------------|---------------|----|
| | NIOZ | n | U. Otago | | | | | |
| Mn (nmol/L) | 0.003 ± 0.001 | 24 | 0.002 ± 0.001 | 18 | | | | |
| Fe (nmol/L) | 0.05 ± 0.02 | 21 | 0.03 ± 0.007 | 17 | | | | |
| Limit of detection (3xSD) | | | | | | | | |
| Mn (nmol/L) | 0,004 | | 0,004 | | | | | |
| Fe (nmol/L) | 0,05 | | 0,02 | | | | | |
| NIOZ | | | | | | | | |
| GEOTRACES South Pacific (GSP) | | | | | | | | |
| Elements | Determined | n | Consensus | Determined | n | Consensus | Determined | n |
| Mn (nmol/L) | 0.787 ± 0.011 | 14 | 0.778 ± 0.034 | 0.421 ± 0.006 | 3 | | 0.443 ± 0.005 | 17 |
| Fe (nmol/L) | 0.150 ± 0.009 | 13 | 0.155 ± 0.045 | 0.722 ± 0.008 | 3 | 0.69 ± 0.04 | 0.597 ± 0.021 | 17 |
| U. Otago | | | | | | | | |
| SAFe D1 | | | | | | | | |
| Elements | Determined | n | Consensus | Elements | Determined | n | Consensus | |
| Mn (nmol/L) | 0.417 ± 0.009 | 6 | | Mn (nmol/L) | 0.121 ± 0.005 | 20 | 0.841 ± 0.019 | |
| Fe (nmol/L) | 0.729 ± 0.018 | 6 | 0.69 ± 0.04 | Fe (nmol/L) | 0.037 ± 0.022 | 19 | 0.774 ± 0.026 | |
| NASS7 | | | | | | | | |
| Elements | Determined | n | Certified | Elements | Determined | n | U. Otago | |
| Mn (µg/L) | 0.72 ± 0.02 | 4 | 0.75 ± 0.06 | Mn (nmol/L) | 0.87 ± 0.12 | 8 | 0.89 ± 0.10 | |
| Fe (µg/L) | 0.336 ± 0.011 | 4 | 0.351 ± 0.026 | Fe (nmol/L) | 0.91 ± 0.62 | 8 | 0.93 ± 0.56 | |
| Station 72 | | | | | | | | |
| NIOZ | | | | | | | | |
| Station 73 | | | | | | | | |
| NIOZ | | | | | | | | |
| U. Otago | | | | | | | | |
| Munida 0 | | | | | | | | |
| Munida + | | | | | | | | |
| Station 72 | | | | | | | | |
| NIOZ | | | | | | | | |
| U. Otago | | | | | | | | |
| Station 73 | | | | | | | | |
| NIOZ | | | | | | | | |
| U. Otago | | | | | | | | |
| Station 73 | | | | | | | | |
| NIOZ | | | | | | | | |
| U. Otago | | | | | | | | |
| Station 73 | | | | | | | | |
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| Station 73 | | | | | | | | |
| NIOZ | | | | | | | | |
| U. Otago | | | | | | | | |
| Station 73 | | | | | | | | |

^a SAFe consensus values were converted to molar units using density value of 1.025 kg/L.

the product of the horizontal eddy diffusivity (K_h , $\text{m}^2 \text{s}^{-1}$) and the curvature of the horizontal DFe gradient ($\frac{\partial^2 \text{DFe}}{\partial x^2}$, $\mu\text{mol m}^{-5}$), as input parameters (de Jong, 2012). The assumed off-shelf value of 0.05 nmol/L was subtracted from the depth-averaged DFe concentration for flux estimates in and below the SML. The concentrations versus distance from the shore for each sampling line were best fitted with a power function. The SML depth was defined as the depth at which the potential density was 0.1 kg/m^{-3} greater than at the surface (Dong et al., 2008) and fluxes below the SML were calculated between the SML and the previously defined permanent pycnocline at a depth of 160 m (Couto et al., 2017).

There is no defined advective velocity from the peninsula towards the off-shelf area for the study region as the main currents flow along the shelf (Fig. 1). Therefore, in this study, horizontal diffusion is used as a plausible mechanism for Fe to be transported from the shelf into the ACC. The horizontal diffusivity (K_h) can be estimated by scale length using the empirical relationship $K_h = 0.0103 \times l^{1.15}$ (de Jong, 2012; Okubo, 1971). The scale length (l) is the distance at which the initial DFe concentration has decreased by 95% (Okubo, 1971) which was estimated as three times the distance by which DFe decreases by $1/e$ of its initial concentration after Johnson et al. (1997). Horizontal diffusivities in the SML along lines 700 and 300 were estimated to be between 360 and $650 \text{ m}^2 \text{ s}^{-1}$. Additionally, horizontal diffusivities were estimated between the SML and the permanent pycnocline resulting in values between 386 and $766 \text{ m}^2 \text{ s}^{-1}$. Horizontal and vertical supplies were estimated for all sampling lines except line 200 as the shelf region of this transect was not sampled. Couto et al. (2017) also pointed out that an anomalously high horizontal diffusivity of $1600 \text{ m}^2 \text{ s}^{-1}$ in the WAP region has been determined by Klinck et al. (2004), but reported this value as unrealistic and used previously reported values of 37 and $200 \text{ m}^2 \text{ s}^{-1}$ for the shelf region (Klinck, 1998; Smith et al., 1999). For consistency among the transects, $548 \pm 135 \text{ m}^2 \text{ s}^{-1}$ (mean ± 1 SD) in the SML and $588 \pm 160 \text{ m}^2 \text{ s}^{-1}$ (mean ± 1 SD) below the SML, the means of the horizontal diffusivities estimated in this study, were instead used to estimate a maximum horizontal supply for each depth interval, and $37 \pm 3 \text{ m}^2 \text{ s}^{-1}$ was used to estimate a minimum horizontal supply both in and below the SML. These values were used throughout rather than using a different diffusivity for each transect to minimize the influence of spatial/temporal differences.

2.4.2. Vertical advective and diffusive supply of DFe

Vertical advection is the product of the upwelling velocity due to Ekman suction (upwelling) and the DFe concentration below the base of the SML. The vertical diffusive supply is the product of the vertical eddy diffusivity and the slope of the DFe depth gradient between the lowest DFe concentration in the SML and below the SML. Grotov et al. (1998) reported vertical advection velocity (upwelling) in the Bellingshausen Sea of between 6.34×10^{-7} and $7.93 \times 10^{-7} \text{ m s}^{-1}$. Furthermore, a large range has been reported for vertical diffusivity (K_z) in WAP waters with a minimum value of below $1 \times 10^{-5} \text{ m}^2 \text{ s}^{-1}$ and an upper value of $1 \times 10^{-4} \text{ m}^2 \text{ s}^{-1}$ (Couto et al., 2017; Hoppema et al., 2002; Howard et al., 2004; Klinck, 1998; Martinson and McKee, 2012; Martinson et al., 2008; Moffat and Meredith, 2018). Therefore, we opted to use both the upper and lower estimates of advection and diffusivity to obtain an upper and lower estimate of vertical DFe supply.

3. Results

3.1. Iron

Generally, the concentration of DFe and TDFe increased towards the continent and with increasing depth. The highest concentrations were measured over the shelf and the lowest concentrations were obtained in the off-shelf SML (Fig. 2A–B). Most stations over the shelf had high concentrations in the SML and subsurface waters (below the SML) (Fig. 3). Additionally, DFe and TDFe concentrations were elevated with

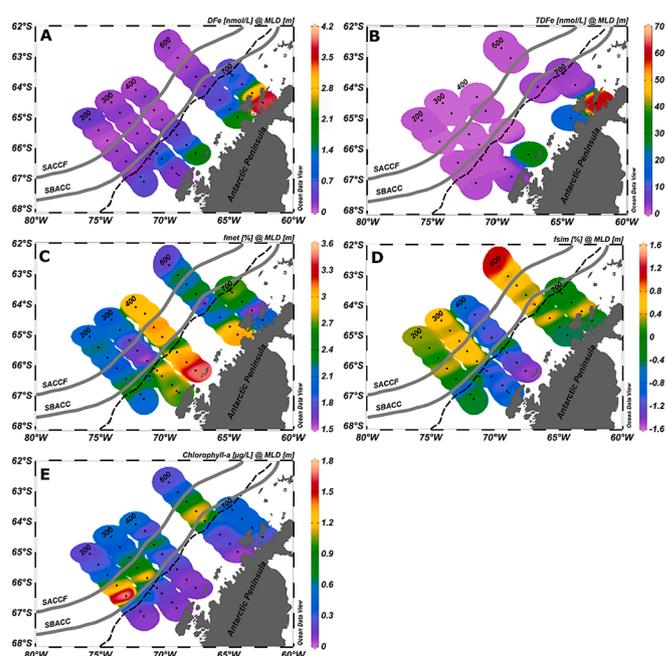


Fig. 2. Color scale surface distributions of depth-averaged Fe, meltwater based on $\delta^{18}\text{O}$, and chlorophyll *a* in the SML showing the highest concentrations between the SACCF and SBACC fronts. SML distribution of A. Dissolved Fe (DFe); B. Total dissolvable Fe (TDFe); C. Sea-ice meltwater fraction (sim); D. Meteoric meltwater fraction (met); E. Chlorophyll *a*. Dashed black line – 500 m isobath shelf break. Abbreviations as in Fig. 1.

depth towards the seafloor (Fig. 3). The Fe concentrations over the shelf ranged from 0.18 – 8.10 nmol/L (mean $1.54 \pm 1.38 \text{ nmol/L}$, median 1.07 nmol/L , $n = 175$) for DFe and 0.40 – 102.34 nmol/L (mean $19.47 \pm 23.82 \text{ nmol/L}$, median 7.95 nmol/L , $n = 84$) for TDFe. Off-shelf Fe concentrations were comparatively low, with values of <0.05 – 1.44 nmol/L (mean $0.38 \pm 0.30 \text{ nmol/L}$, median 0.29 nmol/L , $n = 179$) for DFe and 0.16 – 17.80 nmol/L (mean $2.23 \pm 2.95 \text{ nmol/L}$, median 1.28 nmol/L , $n = 121$) for TDFe.

The highest SML DFe and TDFe concentrations along each sampling line occurred at the stations closest to the continent for sampling lines 700 (station 23), 600 (station 20) and 400 (station 73; Table 2; Fig. 3A–F). The remaining sampling lines 300 and 200 had elevated concentrations of Fe (DFe: 0.31 – 1.84 nmol/L ; TDFe: 0.98 – 2.06 nmol/L ; Fig. 3G–J) at mid-shelf and shelf break stations. However, for sampling line 300, stations close to the peninsula had lower DFe concentrations with only slightly elevated TDFe concentrations (DFe: 0.21 – 0.52 nmol/L ; TDFe: 0.94 – 8.96 nmol/L ; Fig. 3G–H).

Relatively high DFe and TDFe concentrations were also found near the seafloor ($\sim 50 \text{ m}$ above bottom depth) across the continental shelf for all sampling lines. In particular, sampling lines 200 and 300 had significantly higher Fe concentrations near the seafloor (mean DFe: $1.89 \pm 1.30 \text{ nmol/L}$; mean TDFe: $43.62 \pm 30.98 \text{ nmol/L}$) than surface concentrations (mean DFe in MLD: $0.62 \pm 0.47 \text{ nmol/L}$; mean TDFe in SML: $2.05 \pm 2.11 \text{ nmol/L}$). The highest bottom water concentrations were observed along sampling line 300 (stations 70 and 71; Fig. 3G–H). In contrast, along sampling lines 600 and 400, comparatively high DFe and TDFe was observed in the SML (stations 20 and 73; Fig. 3C–F), although bottom water Fe was still elevated (DFe = 2.49 nmol/L and TDFe = 32.14 nmol/L at station 20; DFe = 1.18 nmol/L and TDFe = 17.27 nmol/L at station 73) relative to off-shelf subsurface and deep waters ($>160 \text{ m}$ DFe: $0.52 \pm 0.27 \text{ nmol/L}$; TDFe: $3.51 \pm 4.09 \text{ nmol/L}$). Furthermore, close to the continent, sampling line 700 had similar DFe in the SML and near the seafloor (depth-averaged DFe in SML = 3.05 nmol/L ; DFe near seafloor = 3.40 nmol/L at station 23; Fig. 3A).

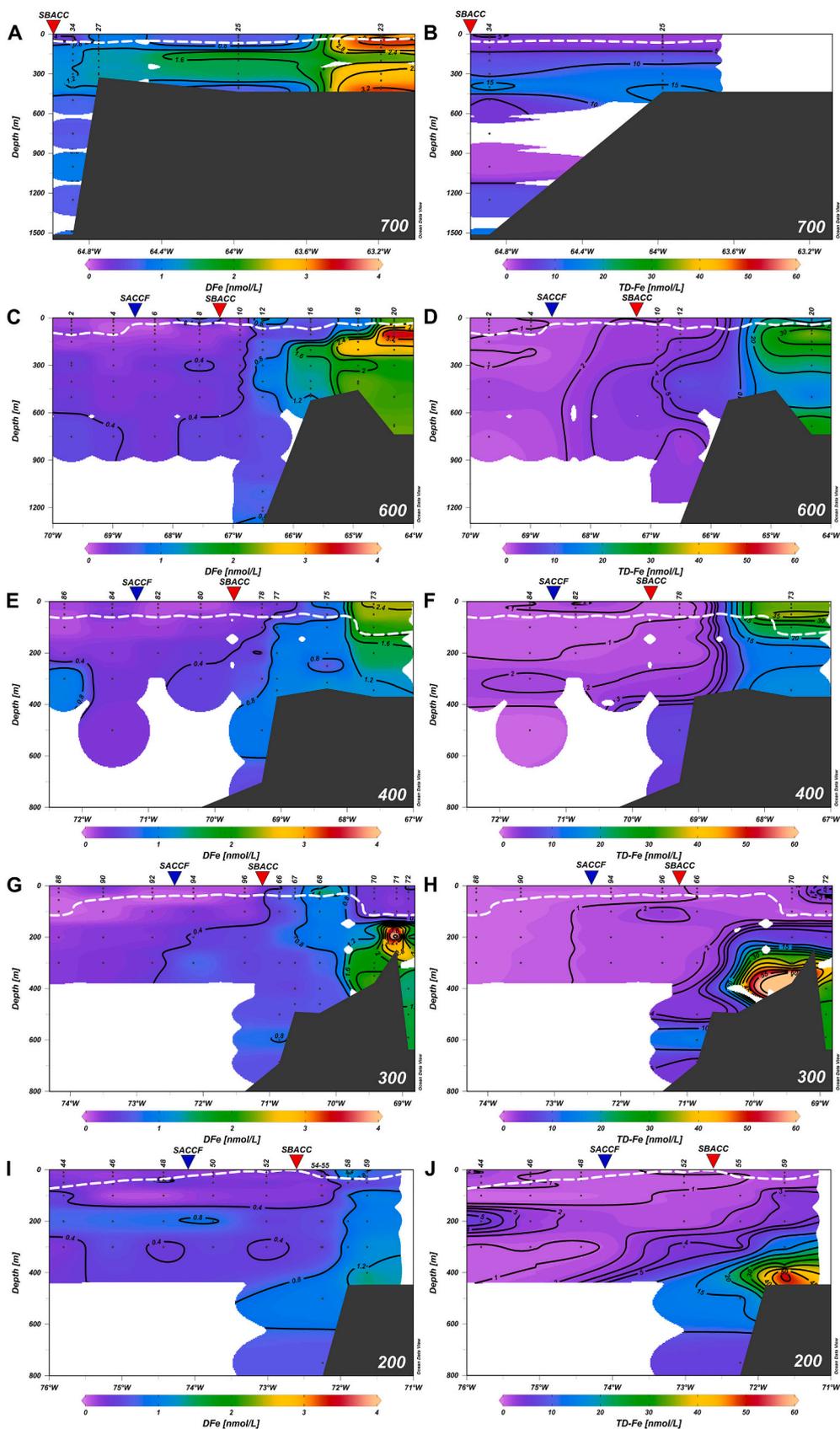


Fig. 3. Distribution of dissolved (DFe) and total dissolvable Fe (TDFe) in color scale along sampling lines 700 (A and B), 600 (C and D), 400 (E and F), 300 (G and H) and 200 (I and J). Please note the color scale is different for DFe and TDFe. Surface mixed layer depth is indicated by a white dashed line and abbreviations as in Fig. 1.

Table 2

Depth-averaged dissolved and total-dissolvable Fe concentrations in and below the surface mixed layer.

| Transect | Station | Bottom Depth | SML | DFe _f _{SML} | TDFe _f _{SML} | DFe _f _{FP} | TDFe _f _{FP} |
|------------------|-------------------------|--------------|-----|---------------------------------|----------------------------------|--------------------------------|---------------------------------|
| # | # | m | m | nmol/L | nmol/L | nmol/L | nmol/L |
| Shelf region | | | | | | | |
| 700 | 21 | 689 | 56 | 3.81 | 59.18 | 3.59 | 42.49 |
| 700 | 23 | 434 | 47 | 3.06 | | 2.86 | |
| 700 | 25 | 435 | 82 | 0.62 | 1.84 | 1.35 | 1.84 |
| 700 | 27 | 326 | 75 | 0.99 | | 1.23 | |
| 700 | 34 | 1512 | 76 | 0.41 | 2.84 | 1.01 | 1.87 |
| 600 | 20 | 736 | 62 | 2.28 | 17.26 | 3.32 | 37.59 |
| 600 | 18 | 455 | 68 | 0.81 | | 1.96 | |
| 600 | 16 | 521 | 85 | 0.49 | | 0.75 | |
| 400 | 73 | 370 | 101 | 2.06 | 29.66 | 1.92 | 20.91 |
| 400 | 75 | 338 | 78 | 0.65 | | 0.94 | |
| 400 | 77 | 370 | 91 | 0.47 | | 1.00 | |
| 400 | 78 | 703 | 79 | 0.17 | 0.54 | 0.25 | 1.23 |
| 300 | 72 | 637 | 106 | 0.26 | 2.77 | 0.55 | 2.22 |
| 300 | 71 | 248 | 108 | 0.19 | | 1.81 | |
| 300 | 70 | 378 | 100 | 0.29 | 1.13 | 0.57 | 2.17 |
| 300 | 68 | 495 | 74 | 1.03 | | 1.00 | |
| 300 | 67 | 490 | 85 | 0.66 | | 0.68 | |
| 300 | 66 | 690 | 76 | 0.46 | 1.02 | 0.36 | 1.55 |
| 300 | 63 | 705 | 85 | 0.21 | 1.20 | 0.28 | 1.31 |
| 200 | 59 | 446 | 75 | 0.45 | 1.50 | 0.96 | 3.57 |
| 200 | 58 | 448 | 87 | 0.84 | | 0.87 | |
| 200 | 55 | 1015 | 95 | 0.15 | 0.66 | 0.18 | 1.24 |
| 200 | 54 | 1660 | 95 | 0.12 | | 0.22 | |
| Off-shelf region | | | | | | | |
| 600 | 12 | 1530 | 78 | 0.68 | 1.74 | 0.50 | 3.46 |
| 600 | 10 | 3073 | 69 | 0.10 | 2.18 | 0.19 | 3.76 |
| 600 | 8 | 3387 | 74 | 0.24 | | 0.11 | |
| 600 | 6 | 3592 | 75 | 0.15 | | 0.08 | |
| 600 | 4 | 3801 | 106 | 0.08 | 0.83 | 0.08 | 0.25 |
| 600 | 2 | 4017 | 127 | 0.15 | 0.77 | 0.09 | 1.23 |
| 400 | 80 | 2500 | 72 | 0.12 | | 0.21 | |
| 400 | 82 | 3431 | 89 | 0.08 | 0.88 | 0.14 | 0.65 |
| 400 | 84 | 3660 | 72 | 0.20 | 0.58 | 0.01 | 0.11 |
| 400 | 86 | 3710 | 100 | 0.09 | | 0.08 | |
| 300 | 96 | 3100 | 50 | 0.16 | 0.50 | 0.17 | 2.16 |
| 300 | 94 | 3564 | 86 | 0.13 | 0.59 | 0.27 | 1.34 |
| 300 | 92 | 3645 | 93 | 0.10 | | 0.12 | |
| 300 | 90 | 3210 | 108 | 0.21 | 0.30 | 0.19 | 0.32 |
| 300 | 88 | 3849 | 119 | 0.11 | 0.20 | 0.10 | 0.20 |
| 200 | 52 | 3692 | 48 | 0.19 | 0.69 | 0.22 | 1.00 |
| 200 | 50 | 3650 | 56 | 0.15 | | 0.30 | |
| 200 | 48 | 3880 | 86 | 0.06 | 0.41 | 0.41 | 0.83 |
| 200 | 46 | 4019 | 101 | 0.18 | 0.80 | 0.44 | 1.21 |
| 200 | 44 | 3990 | 136 | 0.37 | 1.21 | 0.55 | 3.25 |
| Average | Shelf region | | | 0.89 | 9.97 | 1.20 | 9.83 |
| | Off-shelf region | | | 0.18 | 0.84 | 0.21 | 1.41 |
| | Decrease | | | 80% | 92% | 82% | 86% |

3.2. Manganese

Generally, the concentrations of DMn and TDMn increased towards the continent and with increasing depth. The Mn concentrations over the shelf ranged from 0.20–4.14 nmol/L (mean 1.31 ± 0.87 nmol/L, median 1.04 nmol/L, $n = 178$) for DMn and 0.30–5.47 nmol/L (mean 1.84 ± 1.38 nmol/L, median 1.07 nmol/L, $n = 84$) for TDMn. The off-shelf Mn content was relatively low, ranging from 0.08–2.96 nmol/L (mean 0.32 ± 0.34 nmol/L, median 0.24 nmol/L, $n = 189$) for DMn and 0.17–1.76 nmol/L (mean 0.44 ± 0.33 nmol/L, median 0.29 nmol/L, $n = 121$) for TDMn.

3.3. Oxygen isotopes ($\delta^{18}\text{O}$)

Oxygen isotope ratios showed high meltwater input along most sampling lines. The meltwater contribution is a combination of meteoric (glacial melt and precipitation) and/or sea-ice melt. The meteoric fraction increased towards the peninsula and the sea-ice melt contribution increased towards the off-shelf (Fig. 2C–D, S3). A broad tongue,

centering along the sampling line 400, of particularly light water (-0.6 to -0.3‰) extended out from land for the full length of the sampling line. The inshore stations of the sampling lines either side (lines 300 and 600) of line 400 also showed low $\delta^{18}\text{O}$ in the SML, especially so along line 300. The highest meteoric meltwater input to the sampling region occurred along sampling line 400, notably in inner shelf waters (station 73), with the meltwater fraction decreasing off-shelf (Fig. 2D, S3E). The elevated meteoric meltwater fraction was accompanied by deeper mixing (Fig. S2) along this line where the meteoric meltwater fraction was also elevated at depth (Fig. S3E). The lowest meteoric meltwater fraction was estimated for the inner shelf waters of sampling line 700 (station 23) and Gerlache Strait (station 21) (Fig. 2D, S3A). The second highest meteoric influence was estimated for the middle and inner shelf waters of sampling line 300 (Fig. 2D, S3G). The highest positive sea-ice meltwater input was calculated for off-shelf waters and additionally the influence of sea-ice melt was inferred for shelf waters of sampling line 600 (station 16) and inner shelf waters of sampling line 700 (station 23) (Fig. 2C, S3B, S3D). The shelf and off-shelf waters of sampling line 400, as well as the shelf waters of line 300, had a negative sea-ice meltwater

fraction, which is the signature of sea-ice formation (Fig. 2C, S3F, S3H), as was observed in other regions of Antarctic shelf seas (e.g. Meredith et al., 2008; Meredith et al., 2017; Meredith et al., 2013; Randall-Goodwin et al., 2015).

Broadly, there was a similar distribution for meteoric meltwater and Fe in the SML. However, there was no overall correlation between either DFe and TDFe and the sea-ice or meteoric melt fractions in the SML for the whole sampling grid. There was, however, a positive correlation between DFe and the meltwater fractions when assessing the individual sampling transects (Table 3), except for sampling line 200, most likely due to a dearth of samples because of the continuous ice cover preventing sampling towards the continent. However, there was also a positive correlation between DFe and the distance from the coast for the same sampling lines except for line 300 (Table 3). Elevated DFe can be observed in the middle of this transect (station 68) that aligns with observed high-density water in the SML (Fig. S2D). A negative relationship between DFe and the sea-ice fraction was observed for sampling lines 700, 600, 400 and 300 (Table 3).

3.4. Chlorophyll *a*

Chlorophyll *a* concentrations reported previously by Arrigo et al. (2017) are used for the interpretation of the distribution of Fe. The depth-averaged chlorophyll *a* in the SML were elevated between the SACCF and SBACC fronts with the highest concentrations occurring in the southwest of the study region (stations 50 and 52) (Fig. 2E).

4. Discussion

4.1. Historical Fe data in the Pal-LTER grid of the western Antarctic Peninsula

Previously determined near-surface (~2 m depth) DFe for the Pal-LTER sampling grid during the austral summers of 2010, 2011 and 2012 (Annett et al., 2017) showed a larger range in dissolved concentrations than the present study carried out in spring (0.02–8.02 nmol/L, $n = 231$ versus 0.18–4.20 nmol/L, $n = 27$, respectively). Furthermore, lower DFe and TDFe concentrations in the upper 20 m were observed in inner shelf waters adjacent to sampling line 600, southwest of Anvers Island, during the austral summers of 2015 and 2016 (Sherrell et al., 2018), than documented at our station 20 (Table S1, Fig. S4). These observations suggest a stronger surface source influence and/or lower consumption of DFe due to lower phytoplankton abundance during spring. The implicit underlying assumption is that a change in trace metal inventories is directly related to a change in input or removal fluxes. However, it is important to realize that such changes are not necessarily simultaneous, as the inventory is the result of both current and past fluxes. Inherently, there are uncertainties associated with inferring fluxes based on observed concentrations at fixed points in time that thus should be considered as first order indications. In contrast to the surface ocean, elevated subsurface (70–300 m) DFe and TDFe concentrations (Table S1, Fig. S4) were observed during the summer and spring, suggesting similar processes are controlling subsurface Fe in the inner-shelf waters along sampling line 600 during the two seasons.

Table 3

Regressions of DFe concentration vs fraction of meteoric water (%), sea-ice melt (%) and distance (km) in the surface mixed layer.

| Transect | DFe versus | | | | Meteoric fraction | | | | Sea-ice melt fraction | | | | Distance | | | |
|----------|------------|-----------|---------|----|-------------------|------------|---------|----|-----------------------|---------|---------|----|----------|-------|---------|---|
| | r^2 | slope | p-value | n | r^2 | slope | p-value | n | r^2 | slope | p-value | n | r^2 | slope | p-value | n |
| 700* | 0,38 | 0.6 ± 0.3 | 0,03 | 12 | 0,33 | -0.4 ± 0.2 | 0,05 | 12 | 0,67 | -0.02 | | 12 | | | | |
| 600 | 0,48 | 0.9 ± 0.1 | 5E-07 | 41 | 0,42 | -0.7 ± 0.1 | 4E-06 | 41 | 0,58 | -0.003 | | 41 | | | | |
| 400 | 0,29 | 1.3 ± 0.5 | 0,01 | 20 | 0,27 | -0.9 ± 0.4 | 0,02 | 20 | 0,55 | -0.004 | | 20 | | | | |
| 300 | 0,38 | 0.4 ± 0.1 | 5E-04 | 28 | 0,33 | -0.3 ± 0.1 | 0,001 | 28 | 0,06 | -0.0006 | | 28 | | | | |

* Removing anomalous data at the station 23 resulted in a negative slope for the sea-ice fraction relationship ($R^2 = 0.33$, $p = 0.05$) but a positive slope for the meteoric fraction relationship ($R^2 = 0.38$, $p = 0.03$).

Alternatively, seasonal changes do not noticeably affect the Fe inventory between seasons at this depth, which could be related, for example, to a longer residence time of Fe in the deeper part of the water column or the position of the sampling location relative to input sources. In our study, the highest SML Fe concentrations generally occurred in Gerlache Strait (DFe = 4.20 nmol/L, TDFe = 73.64 nmol/L) and in the inner shelf stations of sampling lines 700, 600 and 400 (Fig. 2A–B). Based on these values in the SML, higher DFe and possibly particulate Fe (based on higher TDFe) are found over the shelf during spring rather than summer, most likely due to a combination of sedimentary sources, deep winter mixing, winter sea-ice formation and meltwater influence (Section 4.2.). Surface samples collected in summer from the shelf region (Annett et al., 2017; Sherrell et al., 2018) generally have lower DFe and particulate Fe, but cover a wider range of values than spring samples, implying that DFe is removed unevenly by phytoplankton and/or scavenging by settling particles during the growth season. Additionally, DFe as Fe(III) oxyhydroxides could play an important role, despite the relatively high concentrations of unsaturated iron-binding ligands that were observed in samples from the same expedition, ranging from 0.75 to 4.98 nM eq. Fe (Ardiningsih et al., 2021).

Previously, summer surface DFe concentrations of 0.19 ± 0.28 nmol/L ($n = 71$) were observed seaward of the shelf break (Annett et al., 2017). In the current study, spring SML DFe ranged from <0.05–1.40 nmol/L (mean 0.30 ± 0.33 nmol/L, $n = 26$) and TDFe varied from 0.31–7.21 nmol/L (mean 1.77 ± 1.61 nmol/L, $n = 19$). The mean off-shelf Fe concentrations in spring appear higher than in summer. This is not surprising, as prior to the spring, Fe is introduced into the surface ocean via deep winter mixing (Section 4.4.) and winter sea-ice formation (Section 4.2.3.), while the removal of the Fe by phytoplankton is only just starting (Section 4.6.).

4.2. Surface water sources

Surface mixed layer DFe and TDFe increased towards the peninsula, with the lowest DFe and TDFe occurring in off-shelf waters (Section 3.1). This implies that near-surface Fe sources, such as shallow sedimentary input (Sherrell et al., 2018) or surface sources, such as meteoric and sea-ice meltwater (Annett et al., 2017, 2015), are responsible for supplying Fe to the WAP oceanic region, along with additional input from brine rejection and vertical convection during winter sea-ice formation, atmospheric deposition and/or upwelling of intruded CDW at the glacier front by the meltwater pump. It should be noted here that the meltwater pump, as described for the cavities under the ice shelves of the Amundsen Sea, might not act in a similar fashion for the WAP marine terminating glaciers, and other processes, such as wind-driven upwelling and mixing or vertical eddy diffusion near the glaciers or in the coastal region might play a role too.

4.2.1. Atmospheric deposition

Dust particles/aerosols deposited onto the surface waters of the WAP, either directly or indirectly via meltwater input, could have been transported from Patagonia or the Antarctic continent (Bullard et al., 2016). Given the relatively low aerosol deposition rates in our study region (Gao et al., 2020; Gao et al., 2013; Tagliabue et al., 2009),

atmospheric deposition is deemed to represent only a minor contribution to the Fe concentrations and distributions along the WAP. However, given that aerosols can accumulate over time on sea ice, glaciers and/or icebergs, the contribution to DFe and TDFe from aerosol-derived Fe could increase when meltwater is released into the ocean during ice melting (Duprat et al., 2019; Gao et al., 2020). The indirect contribution to surface Fe content from atmospheric deposition in the WAP region is difficult to distinguish from meltwater input sources and is not discussed further.

4.2.2. Meteoric meltwater

Meteoric water in the study region is dominantly sourced from meltwater runoff, basal ice melting and free-drifting icebergs (Meredith et al., 2017). Land-derived surface meltwater production in the WAP region is relatively large (>50 mm water equivalent y^{-1}), although most of this meltwater refreezes in the snowpack before reaching coastal waters, which reduces the supply of runoff into coastal waters (van Wessem et al., 2017). The rates of glacial discharge (runoff and basal melting) are largest in the regions around Anvers Island (lines 700 and 600), Adelaide Island (line 300) and southern Palmer Land (not in the sampling grid), and lowest in Marguerite Bay (line 200) and in the northern WAP (van Wessem et al., 2017). The $\delta^{18}O$ results of this study show that the input of meteoric water is highest in the inner shelf waters of sampling lines 600, 400 and 300, with much lower meteoric inputs along sampling line 700 (Fig. 2D). Sampling line 700 was the first transect to become ice-free in the austral spring (Annett et al., 2017) and strong currents (Savidge and Amft, 2009) could have diluted the meteoric fraction. In contrast, lines 600, 400 and 300 still had some ice cover during the sampling period (Arrigo et al., 2017) preventing wind stress, and hindering wind-driven currents and mixing. Furthermore, the correlation of DFe with distance landward implies the SML Fe in the study region cannot just be attributed to meteoric melt, as additional shelf processes could also contribute sources of Fe. Notably, the meltwater pump or other coastal mixing processes that bring Fe from deeper waters with associated sediment input would also lead to coinciding increases in both Fe and the meltwater fraction, where only part of the Fe has a meltwater origin (van Leeuwe et al., 2020; St-Laurent et al., 2017; St-Laurent et al., 2019), as would nearby shallow sedimentary Fe sources (Sherrell et al., 2018).

Nevertheless, glaciers, through melting into coastal waters (Annett et al., 2017, 2015), or icebergs melting into the shelf and/or off-shelf waters (Lin et al., 2011), can provide sources of Fe to the Southern Ocean. Notably, basal ice and icebergs can have very high Fe contents due to lithogenic particulate material in the ice, and respective DFe and TDFe as high as 607 nmol/L and 1900 nmol/L in icebergs (Hopwood et al., 2019; Lin et al., 2011), and median values of 0.98 and 108.87 $\mu\text{mol/L}$ in glacial meltwater runoff from the Antarctic Peninsula (Hodson et al., 2017), have been reported. In this study, the DFe of the meteoric meltwater endmember was estimated as 78 ± 38 nmol/L using the regression of DFe versus the meteoric fraction in seawater (Tables 3 and 4), and is an underestimate because conservative behavior of Fe is assumed. This value agrees well with a previous estimate of 66 ± 108 nmol/L for this region (Annett et al., 2017), but is lower than the DFe

Table 4

Regressions of DFe concentration vs fraction of meteoric water (%) in the surface mixed layer and an estimation of the pure meteoric endmember DFe concentration.

| Transect | Slope (nmol/L/ %met) | y-intercept (nmol/L/% met) | r^2 | p- value | DFe in endmember (nmol/L) \pm CI (95%) |
|---|----------------------------|----------------------------------|-------|-------------|--|
| 300 | 0.4 ± 0.1 | -0.8 ± 0.3 | 0,38 | 4E-04 | 41 ± 21 |
| 400 | 1.3 ± 0.5 | -4.0 ± 1.7 | 0,29 | 0,01 | 127 ± 97 |
| 600 | 0.9 ± 0.1 | -1.5 ± 0.4 | 0,48 | 5E-07 | 84 ± 28 |
| 700 | 0.6 ± 0.2 | -1.0 ± 0.7 | 0,38 | 0,03 | 59 ± 53 |
| Mean estimate ($\pm 1SD$) | | | | | 78 ± 37 |

measured directly in local and Antarctic Peninsula glacial meltwater of 348 ± 14 nmol/L (Annett et al., 2017) and 980 nmol/L, respectively (Hodson et al., 2017). This discrepancy is most likely due to the removal of DFe via scavenging, both in the strong salinity gradient upon delivery to the water column as well as during advection to the sampling location, as such high concentrations of Fe cannot be maintained in dissolved form in oxygenated seawater. Overall, meteoric water appears to be an important source of Fe to the study region, but based on currently available data, the exact contribution cannot be quantified with certainty.

4.2.3. Sea-ice

Another source of Fe into surface water is sea-ice melt and winter sea-ice formation. Dissolved Fe in sea ice cores range from 0.23–81.0 nmol/L e.g. (van der Merwe et al., 2011), generally higher than in underlying water (e.g. Lannuzel et al., 2014; Schallenberg et al., 2016).

In this study, a negative correlation between DFe and sea ice melt fraction was generally found in the SML (Table 3), with higher DFe observed in regions with a sea-ice formation signature (negative sea-ice meltwater fraction) (Fig. 2A–C). This indicates that a winter sea-ice formation signal is still present during the spring sampling period of this study, as has also been observed in this and other regions during summer (Meredith et al., 2008; Meredith et al., 2017; 2013; Randall-Goodwin et al., 2015). The elevated DFe and TDFe in regions with a sea-ice formation signature could be due to Fe release into seawater via brine rejection (Schallenberg et al., 2016) or through the resulting vertical convection (Williams et al., 2011) that brings up Fe from underlying waters, especially over shallow parts of the shelf where near-bottom DFe and TDFe levels are high (Fig. 3). The strongest sea ice formation signature was generally found at the inner shelf stations (Fig. 2C) where ice cover is the last to melt, preventing wind-induced mixing and probably preserving the sea-ice formation signature longer. For the remaining shelf waters, the winter sea-ice formation signature is likely diluted due to wind driven mixing and meltwater input from both sea-ice and meteoric meltwater. In contrast to spring, a positive correlation between the sea-ice meltwater fraction and both DFe and particulate Fe was observed in summer (Annett et al., 2017). It appears spring is more dynamic than summer, as the effects of both ice formation and ice melt are generally still detectable, and sea-ice formation, meteoric meltwater, the meltwater pump and/or other coastal mixing processes, are likely to be important sources of Fe to the euphotic zone.

4.2.4. Antarctic Peninsula coastal current (APCC)

A noteworthy feature of the shelf is the formation of the APCC current in the surface water layer during the ice melt period resulting from meltwater runoff (Moffat et al., 2008; Moffat and Meredith, 2018). During our sampling period, the exact location of the APCC could not be defined. However, the APCC is known to traverse the inner shelf waters of sampling lines 700, 600, 400 and 300 (Fig. 1A, yellow line), and the APCC can be identified by a deepening of the mixed layer (e.g. Moffat and Owens, 2008). A deepening to ~ 110 m, indicative of the APCC, was indeed observed for the inner shelf stations (stations 70, 71, 72, 73) of sampling lines 400 and 300 (Fig. 3E–H, S2C–D). Interestingly, SML DFe and TDFe content was low in the inner shelf of sampling line 300, but was elevated at sampling line 400, indicating that DFe input at line 400 does not extend to line 300, or it was transported in a different branch of the current that was not sampled. It seems likely that the APCC current is important for transporting DFe from Anvers Island along Briscoe and Adelaide Islands into Marguerite Bay where it joins the seaward flow out of Marguerite Bay, potentially transporting DFe from the northeast of the shelf towards the southwest and out towards the off-shelf area (Moffat and Meredith, 2018; Savidge and Amft, 2009). Unfortunately, this region was not investigated during our study, but a high surface Fe content in Marguerite Bay during summer could be associated with meteoric meltwater input and/or shallow sedimentary sources

transported within the APCC, in combination with basal melt from the George VI ice shelf. The APCC current could therefore be crucial in supporting productivity in the off-shelf waters seaward of Marguerite Bay, especially during the summer when the current becomes stronger (Moffat and Meredith, 2018; Savidge and Amft, 2009; Marrari et al., 2008; Annett et al., 2017) and should be investigated in future studies. The APCC also flows in the opposite direction to the northeast between the Antarctic Peninsula and Anvers and Brabant Islands through Gerlache Strait towards the South Shetland Islands (Fig. 1A) (Moffat and Meredith, 2018; Savidge and Amft, 2009). In this strait, the highest SML DFe and TDFe of the whole dataset were observed. Below the surface, DFe and TDFe decreased with depth in the upper 100 m, then increased with depth below 100 m (Fig. S8A). The elevated deeper concentrations are most likely related to non-reductive sedimentary input (Section 4.3), whereas the elevated surface concentrations are probably related to the formation of the APCC current and the associated freshwater discharge from the surrounding islands and Peninsula. However, the estimated meteoric meltwater fraction was not high relative to other inner-shelf estimates, implying a range in meteoric meltwater endmembers or a dominant contribution from shallow sedimentary sources (Sherrell et al., 2018). Overall, despite a lack of direct evidence for the role of the APCC in supplying Fe to surface waters in Gerlache Strait, the findings of this study suggest that the APCC could be important for the transport of Fe in this region. However, this interpretation might be biased by the location of the single station in the Gerlache Strait where the proximity to the coastline could also play an important role in surface Fe supply.

4.3. Sedimentary sources

Sedimentary sources play a major role in providing Fe to the shelf waters of the WAP, as is apparent from the elevated DFe and TDFe observed in the bottom shelf waters of the study area (Fig. 3). Sedimentary Fe can be brought up in the water column through diffusion of DFe, the resuspension of sedimentary particles via overflow-driven mixing caused by the bathymetric features of the seafloor of the WAP shelf (Venables et al., 2017), deep shelf break mixing and buoyancy-driven circulation (de Jong, 2012; St-Laurent et al., 2017; Venables et al., 2017) or directly from shallow sediments (Sherrell et al., 2018). According to a modeled Fe budget, approximately 27% of the total Fe input into the shelf region of Antarctica is derived from a sedimentary flux from the extensive continental shelf (Tagliabue et al., 2009). Sedimentary sources of Fe to the shelf region of the study area are especially important for sampling lines 700, 300 and 200 (Fig. 3A–B, G–J) given the relatively high bottom water concentrations at these locations, but elevated near-bottom concentrations are observed along all sampling lines (Fig. 3C–F). High productivity in surface waters can lead to buildup of organic matter in the underlying sediments, resulting in low oxygen conditions and reductive dissolution of Fe (Annett et al., 2017). Due to the faster oxidation and precipitation of Fe relative to Mn in overlying oxygenated seawater, reductive dissolution should lead to a higher effective release of Mn than Fe, and a higher DMn:DFe ratio (Hatta et al., 2013) unless Fe is stabilized by high concentrations of organic ligands present in the water column (Ardiningsih et al., 2021). A higher average inshore DMn:DFe ratio (5.9 ± 3.6 mol:mol) was observed in the more productive near-surface waters of the northern lines than the lower productivity southern lines (1.2 ± 1.2 mol:mol) during the 2011 and 2012 Pal-LTER summer sampling campaigns, and was attributed to reductive sedimentary processes (Annett et al., 2017). During the spring period of the current study, the SML DMn:DFe ratio (1.82 ± 0.61 mol: mol; Table 5) on the shelf was close to the value (1.2 ± 1.2 mol:mol) observed for the lower productivity southern lines of the 2011 summer campaign (Annett et al., 2017). This could imply that high productivity-driven reductive sedimentary processes were not yet operating in spring, or alternatively, that the summer drawdown of DFe drives Mn:Fe ratios to higher values without a seasonal change in DMn.

A signature of reductive sedimentary input on the shelf is more likely

Table 5

Mn:Fe ratio in the study region. Labile particulate (lp) = total dissolvable – dissolved concentration.

| | | | Upper surface layer (<10 m) | SML | Permanent Pycnocline | Below Permanent Pycnocline |
|-----------------|-------|------|-----------------------------|-------------|----------------------|----------------------------|
| Shelf | dMn: | mol: | 1.82 ± 0.61 | 1.94 | 1.68 ± 0.76 | 0.62 ± 0.31 |
| | dFe: | mol: | | ± 0.67 | | |
| | lpMn: | mol: | 0.028 ± 0.015 | 0.034 | 0.034 ± 0.023 | 0.042 ± 0.019 |
| | lpFe: | mol: | | ± 0.027 | | |
| Shelf slope | dMn: | mol: | 2.33 ± 1.58 | 2.10 | 2.01 ± 1.31 | 0.70 ± 0.33 |
| | dFe: | mol: | | ± 1.55 | | |
| | lpMn: | mol: | 0.060 ± 0.030 | 0.053 | 0.060 ± 0.033 | 0.046 ± 0.020 |
| | lpFe: | mol: | | ± 0.031 | | |
| Off-shelf | dMn: | mol: | 0.93 ± 0.53 | 1.12 | 1.24 ± 0.69 | 0.80 ± 0.26 |
| | dFe: | mol: | | ± 0.63 | | |
| | lpMn: | mol: | 0.16 ± 0.18 | 0.27 | 0.22 ± 0.44 | 0.13 ± 0.23 |
| | lpFe: | mol: | | ± 0.49 | | |
| Gerlache Strait | dMn: | mol: | 0.87 ± 0.01 | 0.86 | 0.84 ± 0.06 | 0.36 ± 0.05 |
| | dFe: | mol: | | ± 0.02 | | |
| | lpMn: | mol: | 0.017 ± 0.001 | 0.017 | 0.017 ± 0.001 | 0.016 ± 0.001 |
| | lpFe: | mol: | | ± 0.001 | | |
| Inner Shelf | dMn: | mol: | 1.54 ± 0.32 | 1.63 | 1.32 ± 0.61 | 0.66 ± 0.45 |
| | dFe: | mol: | | ± 0.57 | | |
| | lpMn: | mol: | 0.023 ± 0.010 | 0.026 | 0.027 ± 0.008 | 0.035 ± 0.017 |
| | lpFe: | mol: | | ± 0.013 | | |

to be observed below the permanent pycnocline (160 m, Couto et al., 2017). However, the relatively low DMn:DFe ratio (0.62 ± 0.31 mol: mol; Table 5) on the shelf below 160 m depth demonstrates a higher concentration of DFe than DMn. The presence of such low DMn:DFe ratios contradicts the influence of reductive dissolution of Mn and Fe from sediments on the shelf during the sampling period. Instead of reductive dissolution, a possible mechanism that could lead to the non-reductive sedimentary input of Fe (Homoky et al., 2013) and Mn below the pycnocline is the resuspension of sedimentary particles, followed by desorption from (likely authigenic) phases.

High energy physical processes such as wind-induced wave action (Gargett et al., 2004) and tidal currents (De Jong et al., 2015) can lead to the resuspension of sedimentary particles, forming nepheloid layers in intermediate and bottom waters (de Jong, 2012). Additionally, in the current study region, CDW intrusion through troughs via eddy events (Couto et al., 2017; Martinson and McKee, 2012; Martinson et al., 2008; Moffat and Meredith, 2018; Moffat et al., 2009) or overflow-driven mixing (Venables et al., 2017) could also cause resuspension. Indeed, a nepheloid layer was observed based on relatively low beam transmission close to the seafloor that coincided with increased DFe and TDFe (Fig. S8–9). The shallow ridge near the coast along sampling line 300 could induce transport of DFe from deeper depths into the SML, increasing Fe content, but this was not evident from CTD data as the overlying water column was well stratified (Fig. S2). Moreover, SML DFe and TDFe concentrations overlying the ridge were relatively low, also implying an absence of Fe supply via local upwelling. In contrast, along lines 400, 600 and 700, elevated SML and subsurface (shallower than 200 m) DFe and TDFe were observed over the shelf, but again the strong stratification implies no direct local upwelling (Fig. S2, S7). Some outcropping of near-surface isopycnals can be observed at the shelf break, but strong stratification in subsurface waters at these locations implies that there was no recent deep shelf break mixing (Figs. S2, S6). Most likely, upwelling of deep Fe via the meltwater pump or other mixing processes close to the coast play a role, but given that SML and subsurface DFe and TDFe are generally higher than deep concentrations, an additional source of Fe is required. The latter is most likely meltwater or advection from nearby shallow sedimentary sources.

Based on the current observations, non-reductive sedimentary input of DFe is likely prominent during the spring, and is possibly mediated by ligands (Ardiningsih et al., 2021) or supplied by small colloids (Homoky et al., 2013). This is not surprising given that low phytoplankton growth during spring, inhibited by ice cover, likely results in lower organic matter build-up in shelf sediments and deeper oxygen penetration, whereas higher productivity in summer would lead to higher organic matter and lower oxygen in shelf sediments, facilitating reductive Fe release (Hatta et al., 2013). Interestingly, the breakdown of organic matter can result in both the production of ligands (Ardiningsih et al., 2021) as well as the consumption of oxygen, where different ligands are likely involved in non-reductive solubilization of Fe(III) compared to maintaining Fe(II) in solution from reductive dissolution, which could be the subject of future studies. We suggest that during spring, DFe is introduced from sediments, as the intrusion of CDW in combination with bathymetric features leads to resuspension and desorption of bound Fe from sediment particles, or mixing of porewater into overlying water, increasing bottom water DFe. The influence of sedimentary sources is indeed seen in the increasing DFe in UCDW during its transport landward across the shelf where the initial concentration of 0.37 ± 0.17 nmol/L (mean with 1 SD in off-shelf UCDW) increases to 0.91 ± 0.36 nmol/L (mean with 1 SD in on shelf CDW) in intruded CDW over the shelf that will contribute to SML Fe concentrations after coastal upwelling or mixing.

4.4. Horizontal and vertical Fe supplies

It has been hypothesized that during winter, DFe-rich deep water is entrained into the surface ocean, increasing productivity in the off-shelf waters of the Southern Ocean in the next growth season (Tagliabue et al., 2014). It has also been proposed that winter mixing could supply ten times more Fe than diapycnal diffusion to the surface ocean each year (Tagliabue et al., 2014). During the early austral spring, DFe was low at the off-shelf stations where elevated DFe was observed below the permanent pycnocline, implying that winter mixing was not deep enough to tap into elevated deep Fe sources (Figs. S2 and S5). Thus, significant entrainment of 'deep DFe' is not likely and any DFe present in the SML after winter is removed rapidly, potentially by sea ice algae or by the growing phytoplankton community, as suggested by elevated chlorophyll *a*.

Even though DFe was low in the SML during the spring sampling season, primary producers were active at the off-shelf stations and phytoplankton were not yet suffering from Fe stress (Arrigo et al., 2017; Joy-Warren et al., 2019). Net primary productivity was especially high at the stations located between the SACCF and SBACC fronts (Joy-

Warren et al., 2019), which was attributed to the upwelling of DFe rich waters at this location (Arrigo et al., 2017). However, these elevated productivity levels could potentially also be from a lateral shelf flux of Fe, similar to observations in the offshore regions of Drake Passage (Dulaiova et al., 2009) and the upper mixed layer of the Atlantic Sector of the Southern Ocean (de Jong, 2012).

As discussed previously in Sections 4.2 and 4.3, the shelf waters of the study region had a high inventory of DFe that could be transported from the shelf to the off-shelf region. The estimated supplies (horizontal and vertical) have a wide range due to the range of adopted K_h and K_z values (see Section 2.4) as well as the variable influence of the different Fe sources along the sampling transects. Therefore, we use median supply values per area (inner shelf, mid-shelf, shelf break, continental slope and frontal regions) of the combined transects instead of average values to decrease the influence of the extreme values on the supply estimate (Table 6). As expected, the magnitude of the supply decreased with increasing distance from the peninsula and increasing water column depth, with the highest supplies derived for the inner shelf stations (Table 6). The horizontal supply ($2.7\text{--}40.1 \mu\text{mol m}^{-2} \text{d}^{-1}$; median of all transects for a low and high K_h , respectively) into the SML is higher than the vertical supply ($0.2\text{--}0.3 \mu\text{mol m}^{-2} \text{d}^{-1}$ median of all transects for low and high K_z , respectively) for the inner shelf, which is most likely due to the proximity of sources. Similarly, a stronger influence of the horizontal supply was observed below the SML in inner shelf waters (Table 6). However, with increasing distance from the shore, the vertical supply of Fe becomes higher than the horizontal supply (Table 6). Besides using power functions (see Section 2.5), the horizontal and vertical supplies were also estimated using an exponential function after Gerringa et al. (2012). This also gave larger vertical than horizontal supplies beyond the shelf break and vice versa over the shelf. The Fe inventories and estimated supplies in the SML are compared to the calculated Fe demand (Sections 4.5 and 4.6) to see if the inventory or the combined horizontal and vertical supply is sufficient to support productivity during spring, summer and the whole growing season (spring + summer) over the shelf and beyond the shelf break (off-shelf).

Despite the overall strong attenuation of Fe horizontal supplies towards the off-shelf region, it is evident from the DFe distribution (Fig. 3), that elevated Fe persists further off-shore in the subsurface. This implies that some of the subsurface horizontal Fe supply contributes to the vertical supplies into the SML close to the shelf by generating stronger vertical Fe gradients but this effect becomes negligible in off-shelf waters where elevated Fe was only found below the permanent pycnocline as discussed previously.

The horizontal supply in SML decreased off-shelf, suggesting that continentally derived Fe is largely confined to the shelf and shelf break

Table 6

Estimated DFe supplies in and below the surface mixed layer. SML – surface mixed layer; BSML – below surface mixed layer.

| | Water column | Vertical Supply ^a | | Horizontal Supply ^a | |
|--|--------------|------------------------------|------------------------------|--------------------------------|------------------------------|
| | | Low | High | Low | High |
| | | $\mu\text{mol/m}^2/\text{d}$ | $\mu\text{mol/m}^2/\text{d}$ | $\mu\text{mol/m}^2/\text{d}$ | $\mu\text{mol/m}^2/\text{d}$ |
| Inner Shelf | SML | 0.2 [0.07–0.3] (n = 5) | 0.3 [0.1–0.6] (n = 5) | 2.7 [2.2–4.4] (n = 3) | 40.1 [32.2–65.5] (n = 3) |
| | BSML | 0.1 [0.07–0.3] (n = 5) | 0.2 [0.1–0.7] (n = 5) | 4.5 [2.1–6.1] (n = 3) | 71.8 [33.6–97.1] (n = 3) |
| Mid-shelf | SML | 0.06 [0.06–0.07] (n = 5) | 0.1 [0.1–0.1] (n = 5) | 0.05 [0.04–0.2] (n = 4) | 0.7 [0.5–2.2] (n = 4) |
| | BSML | 0.07 [0.05–0.2] (n = 5) | 0.1 [0.07–0.3] (n = 5) | 0.06 [0.03–0.3] (n = 4) | 1.0 [0.5–4.0] (n = 4) |
| Shelf Break | SML | 0.05 [0.03–0.06] (n = 4) | 0.08 [0.05–0.1] (n = 4) | 0.01 [0.007–0.02] (n = 4) | 0.2 [0.1–0.3] (n = 4) |
| | BSML | 0.07 [0.04–0.1] (n = 4) | 0.1 [0.06–0.2] (n = 4) | 0.02 [0.007–0.02] (n = 4) | 0.3 [0.1–0.3] (n = 4) |
| Continental Slope | SML | 0.04 [0.02–0.07] (n = 4) | 0.07 [0.04–0.2] (n = 4) | 0.006 [0.003–0.01] (n = 4) | 0.08 [0.04–0.2] (n = 4) |
| | BSML | 0.03 [0.02–0.06] (n = 4) | 0.06 [0.03–0.08] (n = 4) | 0.007 [0.004–0.01] (n = 4) | 0.1 [0.07–0.2] (n = 4) |
| SBACC ^b -SACCF ^c | SML | 0.01 [0.008–0.02] (n = 7) | 0.03 [0.02–0.05] (n = 7) | 0.001 [0.0004–0.005] (n = 7) | 0.02 [0.006–0.07] (n = 7) |
| | BSML | 0.02 [0.019–0.03] (n = 7) | 0.04 [0.02–0.05] (n = 7) | 0.002 [0.0005–0.004] (n = 7) | 0.03 [0.008–0.07] (n = 7) |
| NW of SACCF ^c | SML | 0.009 [0.008–0.01] (n = 7) | 0.02 [0.01–0.03] (n = 7) | 0.0004 [0.0003–0.001] (n = 7) | 0.005 [0.005–0.02] (n = 7) |
| | BSML | 0.01 [0.01–0.02] (n = 7) | 0.02 [0.02–0.04] (n = 7) | 0.0004 [0.0003–0.001] (n = 7) | 0.007 [0.005–0.02] (n = 7) |

^a Median values with minimum and maximum values indicated between box brackets and number values (n) calculated for each region in the sampling area.

^b SBACC – Southern Boundary of Antarctic Circumpolar Current.

^c SACCF – Southern Antarctic Circumpolar Current Front.

regions above the permanent pycnocline. This decrease can be observed in the average DFe and TDFe in and below the SML where Fe decreases by 80–90% from shelf to off-shelf waters (Table 2). When productivity increases on the shelf during summer, a further decrease of the lateral Fe flux is expected as a consequence of biological assimilation into, and scavenging onto biogenic particles. Therefore, we deem it likely that continentally derived Fe is largely confined to the WAP shelf region, not only as a result of the local hydrography as suggested previously (Sieber et al., 2021), but also as a result of scavenging and biological uptake.

4.5. Phytoplankton uptake

We observed low DFe in off-shelf SML (<0.1 nmol/L), below the assumed threshold for growth limitation of oceanic phytoplankton (Sedwick et al., 2011), but conversely, relatively high chlorophyll *a* provides no indication of Fe limitation (Arrigo et al., 2017). The parameter Fe^* represents the product of the uptake ratio Fe:N or Fe:P and the nitrate or phosphate concentration subtracted from the DFe (Holmes et al., 2019; Parekh et al., 2005). Depending on the uptake rate of Fe by phytoplankton based on taxonomic and physiological controls on intracellular Fe:C:N:P ratios, one can determine if primary productivity is more likely to be limited by either DFe (negative Fe^*) or the macronutrients nitrate or phosphate (positive Fe^*) (Table S2). This approach is often done to assess the potential for Fe limitation in subsurface waters prior to upwelling, but is used here for the SML to assess the likelihood that this layer will evolve to Fe limitation over time. Either way, this approach has several underlying assumptions: (1) no preferential remineralization of Fe (or nitrate or phosphate, i.e. nutrients remineralize in unison), (2) the inventories observed are representative of the rest of the season with either no additional input, or any additional input is matched by abiotic loss factors, and (3) all DFe is bioavailable and not lost via scavenging (Boyd et al., 2017; Rafter et al., 2017; Richon et al., 2020). When Fe^* is derived using an intermediate Fe:C ratio of 10 $\mu\text{mol}:\text{mol}$ (Twining et al., 2004), nitrate or phosphate is predicted to become a limiting nutrient at the inner shelf stations where there is a high inventory of Fe (Table S2) and DFe is the limiting nutrient everywhere else, as was suggested before (Annett et al., 2017; Sherman et al., 2020). When a low Fe:C uptake ratio of 1.7 $\mu\text{mol}:\text{mol}$ (Twining and Baines, 2013) is adopted, then stations located on the shelf as well as the continental slope are considered to become limited by nitrate or phosphate. Relatively low, but not depleted concentrations of nitrate (2.20 $\mu\text{mol}/\text{L}$) and phosphate (0.10 $\mu\text{mol}/\text{L}$) have been observed during late summer in a coastal bay (Ryder Bay; van Leeuwe et al., 2020), suggesting that major nutrients are not usually limiting in regions near to the major sources of Fe. Given the proximity to Fe sources, other factors such as light limitation most likely prevent full macronutrient depletion (e.g. Twelves et al., 2021). Alternatively, when an extremely high Fe:C uptake ratio of 40 $\mu\text{mol}:\text{mol}$ (Twelves et al., 2021) is used, then DFe is predicted to be the limiting nutrient at all stations. We chose to use different uptake ratios for these calculations, since during early spring, a relatively high uptake ratio is expected, and as the season progresses, the availability of Fe decreases, likely resulting in a decreasing uptake ratio. Such a change in Fe availability could lead to changes in community composition. Indeed, a shift from a *Phaeocystis antarctica*-dominated community to a diatom-dominated community was observed in spring (Arrigo et al., 2017; Joy-Warren et al., 2019), linked to changes in light intensity and possibly a depletion of Fe. The results of this study show that the DFe inventory in off-shelf waters is low from the beginning of the growth season (early austral spring), indicating that most DFe had already been consumed by the phytoplankton community, potentially resulting in Fe limitation in the beginning of the summer season. On the contrary, inner shelf waters have a sufficiently high DFe inventory to potentially deplete the macronutrient inventory during the growth season, whereas mid-shelf phytoplankton communities could only deplete macronutrients by adapting to a lower Fe uptake ratio. However, these estimates are based

on a snapshot in time, and most likely there was a continuous supply of nutrients throughout the season, such as via CDW upwelling or sedimentary supply, as well as remineralization of the nutrients from accumulated biogenic particulate matter.

4.6. Fe demand

Here, we compare Fe supplies with Fe demand from primary productivity, converting satellite-based net primary productivity (NPP) estimates from Joy-Warren et al. (2019), in $\text{mol C}/\text{m}^2/\text{d}$, to Fe demand, in $\mu\text{mol Fe}/\text{m}^2/\text{d}$, using different Fe uptake ratios (see Section 4.5). The underlying rationale is that the combination of calculated vertical and horizontal inputs as the total supply should be sufficient to support productivity during the growth season.

It has been suggested that light is the growth-limiting factor during the sampling period of our study (Arrigo et al., 2017; Joy-Warren et al., 2019; Oliver et al., 2019), and it is likely that Fe limitation develops later in the season, especially off-shelf. Therefore, here we assess if the horizontal DFe supply into the SML from the shore to the off-shelf, and/or the vertical supply from underlying waters into the SML is sufficient, either separately or combined, to support the estimated NPP throughout the season (Table 7), noting that these are first order supply estimates with large uncertainties (see Section 2.4). As detailed in Section 4.4, we used the median value of the estimated supplies per area, calculated using the upper and lower limits for diffusivity and advection coefficients.

Our estimated supply resulting from the horizontal flux to the entire sampling area (0.005–0.07 $\mu\text{mol m}^{-2} \text{d}^{-1}$, assumed to be constant during the season) is not enough to support average NPP for all growth over the entire season in the region under moderate (Fe:C ratio of 10 $\mu\text{mol}:\text{mol}$) or high (Fe:C ratio of 40 $\mu\text{mol}:\text{mol}$) uptake ratios. Even with a low uptake ratio (Fe:C ratio of 1.7 $\mu\text{mol}:\text{mol}$), average NPP can only be supported with high diffusivity coefficients (Table 7). The horizontal supply to off-shelf waters (0.001–0.02 $\mu\text{mol m}^{-2} \text{d}^{-1}$), is only sufficient to support average NPP if a high diffusivity coefficient is assumed and when using a low uptake ratio (Fe demand of $0.015 \pm 0.006 \mu\text{mol m}^{-2} \text{d}^{-1}$). In contrast, horizontal supply over the shelf (0.05–0.7 $\mu\text{mol m}^{-2} \text{d}^{-1}$) should be sufficient to support NPP under any uptake ratio (Table 7). These results again suggest that horizontal supply within the SML might not be an important source of the limiting micronutrient Fe for the off-shelf region.

The calculated supply via the vertical flux (0.03–0.05 $\mu\text{mol m}^{-2} \text{d}^{-1}$, assumed to be constant during the season) is only sufficient to support the average estimated NPP under a low uptake ratio for all regions and seasons (Table 7). The estimated supply of the total flux (sum of horizontal and vertical supply) is only high enough to support average NPP under low uptake ratios in the off-shelf region and under low, moderate, or high uptake ratios in the shelf region. These supply estimates might only be representative of the early spring, as even lower SML Fe concentrations occur during summer, with relatively invariant deep water concentrations, implying a stronger vertical gradient in summer that will enhance vertical supply. Nevertheless, these estimations imply that the phytoplankton community has a relatively low average uptake ratio that is likely between 1.7 and 10 $\mu\text{mol}:\text{mol}$ Fe:C in the off-shelf region and in the range of 10 to 40 $\mu\text{mol}:\text{mol}$ Fe:C for the shelf region during the sampling period of this study.

The large range and variation in the supply of Fe between different locations (Table 5) supports the hypothesis of patchy Fe limitation in WAP shelf waters (Garibotti et al., 2003; Prézelin et al., 2004; Smith et al., 2008) as well as heterogeneous Fe inputs (Annett et al., 2017). Additionally, the results of our study suggest that the horizontal flux has a greater influence on the supply of Fe in the shelf region than in the off-shelf region (Section 4.4.), which is not surprising given the proximity of Fe sources, such as glaciers and shallow sediments. In contrast, the influence of the vertical flux is dominant in the off-shelf region (Section 4.4.).

Table 7
Net primary productivity (NPP) and estimated DFe demand. a, b, c

| | NPP mol C/m ² /d | Fe Demand | | | Vertical ^a Supply μmol/m ² /d | Horizontal ^a Supply μmol/m ² /d | Total Supply ^b | |
|---|--------------------------------|------------------------|------------------------|------------------------|---|---|----------------------------|--------------------------|
| | | Low | Medium | High | | | Low | High |
| | | μmol/m ² /d | μmol/m ² /d | μmol/m ² /d | | | μmol/m ² /d | μmol/m ² /d |
| Sampling area during entire growth season | 0.011 ± 0.004 | 0.018 ± 0.007 | 0.11 ± 0.04 | 0.42 ± 0.17 | 0.03–0.05 | 0.005–0.07 | 0.03 [0.008–4.6] (n = 29) | 0.1 [0.02–65.8] (n = 29) |
| Off-shelf region during entire growth season | 0.009 ± 0.003 | 0.015 ± 0.006 | 0.09 ± 0.03 | 0.36 ± 0.14 | 0.01–0.02 | 0.001–0.02 | 0.01 [0.008–0.07] (n = 18) | 0.04 [0.02–0.3] (n = 18) |
| Shelf region during entire growth season | 0.018 ± 0.010 | 0.03 ± 0.02 | 0.18 ± 0.10 | 0.73 ± 0.40 | 0.06–0.1 | 0.05–0.7 | 0.1 [0.04–4.6] (n = 11) | 0.8 [0.2–65.8] (n = 11) |
| Sampling area during spring season ^c | 0.010 ± 0.003 | 0.017 ± 0.005 | 0.10 ± 0.03 | 0.40 ± 0.12 | 0.03–0.05 | 0.005–0.07 | 0.03 [0.008–4.6] (n = 29) | 0.1 [0.02–65.8] (n = 29) |
| Sampling area during summer season | 0.013 ± 0.004 | 0.022 ± 0.008 | 0.13 ± 0.04 | 0.53 ± 0.18 | 0.03–0.05 | 0.005–0.07 | 0.03 [0.008–4.6] (n = 29) | 0.1 [0.02–65.8] (n = 29) |

^a Median values of vertical and horizontal supplies based on a high and low K_z and K_h .

^b Median values with minimum and maximum values indicated between box brackets and number of values calculated for each region in the sampling area.

^c Same vertical and horizontal supplies for spring and summer seasons.

5. Conclusions

In the present study, samples were collected across a range of water depths, in order to assess the depth distributions as well as the surface concentrations of DFe and TDFe during the austral spring in one of the regions of Antarctica that is being strongly impacted by changing climate. In the WAP oceanic region, elevated meteoric meltwater fractions and the presence of a sea-ice formation signal coincided with high DFe and TDFe in the SML. Deeper in the water column, the benthic release of DFe (e.g. resuspended sedimentary particles and/or porewaters) was inferred, suggesting an important role for sediment resuspension as a result of CDW intrusion and overflow mixing facilitated by the bathymetric features of the shelf. The concentrations of DFe and TDFe in spring appear to be mainly controlled by the influence of sea-ice formation during the preceding winter, and the supply of Fe from meteoric meltwater and from below via coastal upwelling or mixing, or supply from nearby shallow sediments. To improve understanding of the Fe distribution and the potential for horizontal and vertical transport, higher resolution sampling is needed to constrain the influence of small-scale features, such as intruding eddies, and heterogeneous sources. Additionally, characterization of the sedimentary pore-water biogeochemistry during spring and summer could reveal if indeed Fe is mainly released via non-reductive dissolution or if there is a seasonal progression towards reductive dissolution, and how this might change in this rapidly changing environment.

Antarctic Peninsula glaciers have undergone significant retreat due to an increase in surface air temperature and ocean forcing, primarily via the intrusion of relatively warm CDW waters onto the shelf (Cook et al., 2016). Antarctic ice sheet retreat has been suggested to result in an increase in lateral Fe supply from meltwater (Death et al., 2014; Wadham et al., 2013) but input from shallow sediments combined with coastal upwelling or mixing has been suggested as another, and potentially even more important source of Fe into the surface water (e.g. St-Laurent et al., 2019; Twelves et al., 2021). Estimated DFe supplies for the current study region suggest that the horizontal supply into the SML is higher than the vertical supply for shelf stations. However, the vertical supply becomes more prominent seaward where it becomes the largest supply, notably between SBACC and SACCF as suggested before (Arrigo et al., 2017). The majority of DFe and TDFe introduced into shelf waters is removed in the shelf region prior to reaching off-shelf waters in either the SML or the subsurface. Consequently, the calculated horizontal supply is currently estimated to be insufficient to support the observed net primary productivity in the off-shelf region. Thus, a future increase in Fe input might not have profound consequences for Fe availability in off-shelf waters. Furthermore, the currently estimated total supply into

the off-shelf SML is barely enough to support observed net primary productivity throughout the growth season. This is opposite to the shelf region where the total flux supplies enough Fe to maintain observed NPP, even when assuming a high Fe:C uptake rate. If the trend of a decrease in sea-ice cover continues, this could result in a lower supply of Fe from below in the off-shelf region. Such a decrease in the vertical flux would be due to a smaller region of winter entrainment that is driven by brine rejection during sea-ice formation that deepens the mixed layer depth (Tagliabue et al., 2014). Additionally, increasing melt rates and water temperatures might result in stronger stratification in the surface layer and weaker vertical diffusivity between the surface water layer and the water beneath the pycnocline (Sarmiento et al., 2004) but the consequences of stronger stratification on nutrient supply from below are subject to ongoing discussion (van Haren et al., 2021). Overall, predictions about the future of iron cycling and its effect on primary productivity in WAP waters and the Southern Ocean remain uncertain, and it is abundantly clear that many intricacies of this unique and interconnected system remain unresolved.

Declaration of Competing Interest

None.

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Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.marchem.2021.104066>.

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