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Fish Tropomyosin and Fish Nucleotropomyosin

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Tropomyosin has been prepared from muscles of several animals (rabbit, pig, horse and whiting) by dehydrating the minced muscle with organic solvents and extracting with α -potassium chloride (Bailey, 1948). A detailed account of its isolation and properties has also been published (Bailey, 1948; Bailey, Gutfreund & Ogston, 1948; Astbury, Reed & Spark, 1948).

In the course of an extensive ultracentrifugal study on fish myosins, it has been found that two proteins showing the same solubility properties as tropomyosin can be isolated from fish muscles without previous treatment with organic solvents. Both tropomyosins have been obtained in crystalline form and their ultracentrifugal and electrophoretic behaviours have been investigated. One of these proteins is practically identical with Bailey's tropomyosin. Dr K. Bailey kindly suggested that the other, which had the larger sedimentation constant, might be an association of the first compound and nucleic acid. This was completely confirmed by phosphorus and ribose determinations and by the ultraviolet absorption. The name 'nucleotropomyosin' is proposed for this compound.

EXPERIMENTAL

Isolation

Extraction of fish muscles with salt solutions (mixtures of phosphate and potassium chloride) of varying pH between 5 and 7 shows that the pro-

portion of tropomyosin in the extract increases as the pH of the extraction fluid decreases and the time of extraction is short. The tropomyosin is still very extractable when the duration of extraction is 10 min. and the pH of the extract is 5. The amount of myosins going into solution under these conditions becomes much smaller. The proportion of tropomyosins in the extracts can be increased if the extraction is carried out with water kept for several months in the frozen state. This treatment decreases the extractability of myosins (Reay & Kuchel, 1936) and does not alter the behaviour of the tropomyosins. Separation of nucleotropomyosin from tropomyosin in the extracts is easy. On dilution with water, nucleotropomyosin coprecipitates with the myosins while tropomyosin remains in solution and precipitates only on the supernatant to pH 4.6. A final purification of the substances is carried out by ammonium fractionation. The details of the method are given below.

The preparation has been carried out through the use of cold and all the separations have been done by centrifugation. Carp muscles cut with a freezing microtome into thin slices (Dubuisson, 1947) are extracted for 20 min. in a KCl-phosphate solution (0.15 M-KCl and 0.05 M-phosphate, pH 5). This extract is diluted with 3 vol. of water; the precipitate is discarded and the supernatant is kept.

The residue from this first extraction is re-extracted for 10 min. with 3 vol. of 0.5 M-phosphate solution containing 0.3% sodium adenosinetriphosphate.

residue is discarded and the extract (II) is diluted with 7 vol. of water. A precipitate of myosins and nucleotropomyosin forms which is washed twice with water and redissolved in 0.5 M-KCl at neutral pH. The supernatant (II) and the supernatant (I) are mixed and brought to pH 4.6. The precipitate containing tropomyosin is washed twice with water and redissolved in 0.5 M-KCl at neutral pH.

Both solutions are now centrifuged for 30 min. at 14,000 rev. min. to remove some turbid material and are purified by a second precipitation by dilution with 8 vol. of water at neutral pH (nucleotropomyosin) or at pH 4.6 (tropomyosin). Both precipitates are washed twice with water and redissolved in 0.5 M-KCl at neutral pH.

Both tropomyosins are isolated from these solutions by $(\text{NH}_4)_2\text{SO}_4$ fractionation at neutral pH: the major part of the total protein content of the solutions precipitates between 30 and 50% saturation, whilst the tropomyosins precipitate between 50 and 66% saturation. The precipitate can be redissolved very quickly by a slight dilution with water, giving a water-clear solution.

Although these methods of preparation are very reproducible, some variations are observed in the yields obtained, which are usually about 0.07% of the wet weight for nucleotropomyosin and 0.03% for tropomyosin. A better yield, amounting to 0.5% of the fresh muscle weight, has been obtained after organic solvent treatment (Bailey, 1948). This discrepancy is not due to loss during the isolation. The quantity of tropomyosins and myosins extracted from fresh muscles under the conditions described amounts to 1% of the wet weight and the tropomyosin content of the solution varies from 10 to 20%. Fractionation by $(\text{NH}_4)_2\text{SO}_4$ is very effective: no precipitation of tropomyosin occurs before 40% saturation while the major part of the myosins has already been removed. The denaturation of the myosins by organic solvent treatment probably increases the differences in extractability of the myosins and tropomyosins and allows a more complete removal of the tropomyosins.

Crystallization. The undiluted salted-out precipitates are used for crystallization. This is carried out using the conditions previously described for tropomyosin (Bailey, 1948), i.e. by dialysing an approx. 1.5% solution against a solution containing 16 g. $(\text{NH}_4)_2\text{SO}_4$ l. and 0.01 M-acetate buffer of pH 5.4. Nucleotropomyosin crystallizes in elongated prisms (Figs. 1, 2); tropomyosin in the quadrangular plates previously described by Bailey (Fig. 3). From a private communication of Dr Bailey it would appear that the crystalline form of nucleotropomyosin is not characteristic of this compound alone; similar crystals have already been obtained by him from fish (whiting) tropomyosin solutions prepared in the usual way.

Ultracentrifugation

The solutions were examined in the Svedberg oil-turbine ultracentrifuge using the Philpot optical system (Philpot, 1938) at 60,000 rev. min. The tropomyosins were dissolved in a phosphate buffer of ionic strength 0.1 containing 0.25 M-sodium chloride. The pH of the solution was 7.1. The concentrations of the protein were determined refractometrically, assuming a specific refractive increment of 0.00180. The values of the sedimentation constants have been corrected to 20° and to a water basis, using the

partial specific volume given by S. Pedersen (1940), Appendix II.

The preparation of tropomyosin appears homogeneous by ultracentrifugation. It sediments as a single fairly symmetrical peak whatever concentration (Fig. 4*a*), and the values of the friction constants at different concentrations from different preparations are in good agreement (Fig. 5 and Table 1). The sedimentation constant extrapolated to zero concentration is 2.8 S values previously obtained for rabbit trypsin (Bailey *et al.*, 1948) have been added to them; they show excellent agreement with the determinations.

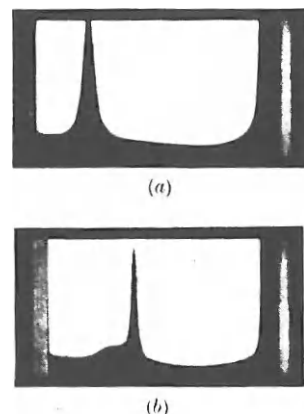


Fig. 4. Sedimentation diagram of tropomyosin (a) and nucleotropomyosin (b). Migration to the right. Tropomyosin: 0.89%, 100 min. after full speed. Nucleotropomyosin: 0.07%, 90 min. after reaching full speed.

The behaviour of the preparation of nucleotropomyosin is more complicated: it sediments as a sharp peak which spreads much more slowly than tropomyosin and is always followed by a small shoulder (Fig. 4*b*). The proportion of this shoulder increases at low concentrations; it is not a dissociation product of the main component but a sedimentation constant-concentration product which (Fig. 5) cannot therefore be extrapolated to zero concentration.

Several ultracentrifugations have been carried out to check the stability of nucleotropomyosin. The sedimentation is unchanged in the presence of adenosine triphosphate (ATP) or after keeping the prepared solution at the pH of crystallization for a few days. In acetate buffer of pH 3.5 nucleotropomyosin is slowly and irreversibly transformed (as shown in Fig. 6) a new component sedimenting more slowly appears after 2 hr. (Fig. 6*a*), becomes prominent after 20 hr. (Fig. 6*b*). Tropomyosin as an entity disappears completely if the solution is kept a longer time at this pH. Sedimentation runs made with solutions of nucle-

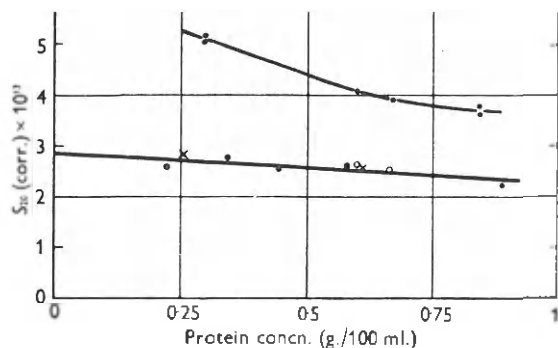


Fig. 5. Variation of S_{20} for tropomyosin and nucleotropomyosin with the concentration of the solution. Lower curve: tropomyosin; upper curve, nucleotropomyosin. \circ , values taken from Bailey *et al.* (1948); \times , values obtained by transformation of nucleotropomyosin at low pH.

Table 1. Sedimentation data for tropomyosin and nucleotropomyosin at different concentrations in acetate buffer (I = 0.35, pH 7.1.)

Tropomyosin			Nucleotropomyosin		
Sample no.	Protein concn. (g./100 ml.)	S_{20} corr. $\times 10^{12}$	Sample no.	Protein concn. (g./100 ml.)	S_{20} corr. $\times 10^{12}$
1	0.58	2.57	1	0.84	3.79
2	0.34	2.73	2	0.84	3.63
3	0.89	2.22		0.6	4.08
	0.44	2.52		0.3	5.11
	0.22	2.59	3	0.67	3.95
4*	0.25	2.82		0.3	5.14
5*	0.61	2.57			

* Obtained by transformation of nucleotropomyosin at pH 3.5.

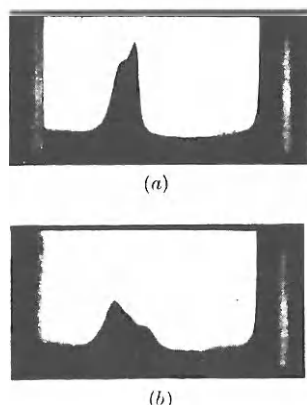


Fig. 6. Sedimentation diagrams of nucleotropomyosin kept in acetate buffer pH 3.5 for 2 hr. (a, total concn. 1.1 μ g) and 20 hr. (b, total concn. 0.87 μ g); both 130 min. after reaching full speed. Migration to the right.

myosin kept 20 hr. or more in acetate buffer at pH 3.5 show that the peak corresponding to this new component has the same shape and rate of sedimentation as tropomyosin (Fig. 5 and Table 1). The

tropomyosin are transformed at low pH to myosin.

Electrophoresis

The experiments were performed with the electrophoresis apparatus slightly modified by Dubuisson & Jacob, 1945; Dubuisson, Debot, 1950). They lasted about 20 hr. at $I = 0.15$ (3.6 V./cm.) and 7 hr. at $I = 0.15$ (3.6 V./cm.) giving a migration of approx. 6 cm. in both directions. No account was taken, in the calculation of mobilities, of the differences in conductivity between the dialysis fluid and the protein solution, the same dialysis fluid being used for both limbs.

Both preparations were homogeneous as shown by electrophoresis, showing a single component with only a slight admixture of impurities. No difference in mobility was observed between nucleotropomyosin and tropomyosin.

myosin and tropomyosin in several experiments made on pure solutions and on mixtures of the two. The mobilities found are given in Table 2.

Table 2. Electrophoretic mobilities of tropomyosin and nucleotropomyosin

Conditions of electrophoresis	Mobility (10^{-5} sq. cm./sec.)
NaCl-phosphate buffer of I = 0.35 and pH 7.1 (I = 0.1 for phosphate and 0.25 for NaCl)	-4.30
NaCl-phosphate buffer of I = 0.15 and pH 7.4 (I = 0.1 for phosphate and 0.05 for NaCl)	-6.40
NaCl-acetate buffer of I = 0.15 and pH 3.5 (I = 0.1 for acetate and 0.05 for NaCl)	+6.15*

* Approximate value; one determination only.

The values obtained at $I = 0.35$ and pH 7.1 are greatly different from the mobilities of rabbit tro-



Fig. 1. Nucleotropomyosin
(magnification $\times 54$).



Fig. 2. Nucleotropomyosin
(magnification $\times 135$).



Fig. 3. Tropomyosin (magnification $\times 135$).

under the same conditions (ascending, 5-6; descending, 4-9; Duboussin, 1950). Further investigations will be necessary to determine the cause of this difference.

Nucleic acid content

Liberation of the nucleic acid was performed by heating the protein solution in a boiling-water bath in 5% (w/v) trichloroacetic acid for 45 min. The protein was removed by filtration and the ribose content of the filtrate was determined according to Meijbaum (1939), using a 45 min. heating time as proposed by Albaum & Umbreit (1947). Measurements were made with a photoelectric absorptiometer using the 607 Ilford filter with a maximum transmission at 600 m μ . Readings were referred to a calibration curve obtained by measuring the absorption of different amounts of D-ribose under the same conditions. As both deoxyribose and ribose react with the orcinol reagent of Meijbaum (1939), the diphenylamine reaction specific for deoxyribose was performed on the trichloroacetic filtrate under the conditions described by Dische (1930). The total phosphorus contents of nucleotropomyosin and tropomyosin were determined by the perchloric acid method as modified by Allen (1940), using the same photoelectric absorptiometer as for ribose and the 608 Ilford filter.

The ultraviolet spectrum of both proteins dissolved in the phosphate-sodium chloride buffer of pH 7.1 and $I=0.35$ was measured against the solvent with the Beckman universal spectrophotometer (model D.U.). (These solutions had already been used in the ultracentrifugal experiments.) Readings were taken at 3 m μ . intervals between 220 and 350 m μ . The concentrations of the solutions, expressed in g./100 ml., were determined by micro-Kjeldahl assuming a nitrogen content of 16.6% (Bailey, 1948) and were used for the calculation of the extinction coefficient $E_{1\%}^{1\text{cm}}$.

Both preparations were precipitated by an equal volume of 10% (w/v) trichloroacetic acid and kept in a boiling-water bath for different intervals to determine the time required for complete extraction (Table 3). (A slight liberation of nucleic acid occurs even after precipitation at room temperature.) Approximately constant values were obtained after about 20 min. for tropomyosin and 30 min. for

nucleotropomyosin. A heating interval of 45 min. was therefore chosen for the determinations.

Table 3. Rate of nucleic acid liberation at 100°C. in 5% (w/v) trichloroacetic acid

Duration of heating (min.)	Ribose liberated (mg./g. prote)	
	Tropomyosin	Nucleotropomyosin
0	4.4	5.35
10	10.2	32.1
18	10.4	—
24	—	34.6
30	12.7	44.0
45	11.0	44.0
60	10.0	—

The ribose and phosphorus contents found are given in Table 4. In view of the very different liberation curves given by ribose and deoxyribose with the orcinol reagent (Schneider, 1945) and the agreement observed between the ribose and phosphorus determinations, the presence of deoxyribose is very unlikely. Tests for deoxyribose carried out on both preparations of nucleotropomyosin fail to give any evidence of the presence of this sugar.

Samples of the first preparation were used for measurements of the ultraviolet absorption (Fig. 7).

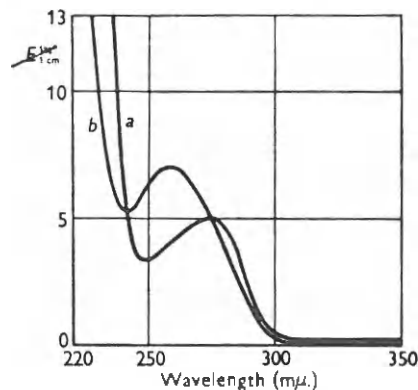


Fig. 7. Ultraviolet absorption spectra of tropomyosin (curve a) and nucleotropomyosin (curve b) in a phosphate-sodium chloride buffer of pH 7.1 and $I=0.35$. Ordinate: $E_{1\%}^{1\text{cm}}$ for tropomyosin and 0.25 $E_{1\%}^{1\text{cm}}$ for nucleotropomyosin.

Nucleotropomyosin shows the well known absorption maximum at about 260 m μ , and tropomyosin a maximum at 274 m μ due to its content

Table 4. Ribose and phosphorus contents of tropomyosin and nucleotropomyosin

		Ribose (%)	P (%)	Ribose/P (mol./atom)	Pentose acid content (%)
Preparation I	Tropomyosin	1.05	—	—	2.5
	Nucleotropomyosin	4.38	—	—	9.5
Preparation II	Tropomyosin	1.22	0.20	1.25	2.5
	Nucleotropomyosin	6.75	1.47	0.99	15.5

Calculated from the ribose content.

